48. A SUMMARY OF INTERSTITIAL-WATER GEOCHEMISTRY OF LEG 1331

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ABSTRACT

The chemical compositions of the interstitial pore-water samples measured on board the *JOIDES Resolution* are compared with shore-based analyses of the samples' strontium, oxygen, and carbon isotopic compositions. These analyses support conclusions of large-scale water movement through the sediment overlying the Queensland Plateau at Sites 811, 812, 813, and 814, possibly leading to massive platform dolomitization.

At the Queensland continental margin sites, complete sulfate reduction and abundant methanogenesis occurs, yet little evidence of these processes is seen in the $\delta^{13}C$ of the pore waters. The failure to observe these signatures is a consequence of large-scale recrystallization of carbonate sediments masking the isotopic signatures of these processes. Although abundant igneous material is present at these sites, we do not think that significant alteration of these minerals is taking place in situ. Decreases in the Mg^{2+} and Ca^{2+} concentration result from the formation of dolomite and calcite; decreases in K⁺ are a consequence of absorption onto clays; and decreases in the $\delta^{18}O$ are caused by carbonate recrystallization, with perhaps some minor influence from the alteration of volcanic glass.

At several sites there is evidence of diffusion of saline fluids from an underlying source, presumably evaporitic in nature, but this cannot be confirmed using either Sr or O isotopic data.

INTRODUCTION

During Leg 133 of the Ocean Drilling Program (ODP), sediments were cored in three different areas off the northeastern coast of Australia. All of these areas contained periplatform sediments (Schlager and James, 1978). This type of sediment is generally found near shallow-water carbonate platforms and consists of mixtures of metastable minerals, high-Mg calcite (HMC), and aragonite, as well as low-Mg calcite (LMC). The first area includes those sites (811, 812, 813, 814, 816, 817, 818, 824, 825, and 826) situated on submerged carbonate platforms or on the flanks of such platforms: the Queensland and the Marion plateaus (Fig. 1). The mineralogy of the sediments at these sites is predominantly carbonate, containing only small concentrations of noncarbonate minerals such as quartz and clay minerals. The second group of sites (815 and 823) is situated in the deeper water troughs between the Queensland and the Marion plateaus and the Great Barrier Reef (Fig. 1). These sites penetrated thick sequences of mixed carbonates (predominantly LMC) and siliciclastics. The third group (Sites 821, 820, 819, and 822) forms a proximal-to-distal transect from the Great Barrier Reef into the adjacent Queensland Trough (Fig. 1). This third group is situated in shallower water than is the second group and contains substantial concentrations of periplatform sediments mixed with quartz, feldspar, and clay minerals.

The aim of this paper is to synthesize the shipboard chemical data with oxygen and carbon isotopic analyses of pore waters, and to integrate them with data presented by Swart (this volume) and Elderfield et al. (this volume).

METHODS

The chemical data discussed in this paper have been presented in other papers, and, therefore, the reader is referred to these references for the appropriate analytical techniques: Davies, McKenzie, Palmer-Julson, et al. (1991); Elderfield et al. (this volume); and Swart (this volume). The C and O isotopic data for the pore waters, which have not been presented previously, were obtained using the methods described in Swart (this volume).

RESULTS

A summary of all geochemical data, including the Sr, C, and O isotopic data as well as measured shipboard data (with the exception of ammonia, phosphate, and silica), is shown in Table 1. As a result of some analytical problems, C and O isotopic data are not available for all samples.

Oxygen Isotopic Data

Sites 811, 812, 813, and 816

These sites are characterized by relatively heavy oxygen isotopic compositions in their pore waters throughout the cores (Fig. 2 and Table 1). Generally, these isotopic values lie between +0.5% and +1.0%. Site 813 exhibits slightly depleted isotopic values relative to Sites 811 and 812, with δ^{18} O values lying between +0.2‰ and +0.5‰.

Sites 819, 820, 821, 822, and 823

These sites show an enrichment in their δ^{18} O values in the upper 50 to 100 mbsf, followed by a depletion down the core. The magnitude of this depletion increases with distance from the continental margin. The most depleted δ^{18} O values of -0.97% are reached in the sediments of Site 822 at a depth of 397.50 mbsf (Fig. 3 and Table 1). Site 823, which is actually the farthest away from the continental margin, reaches a minimum of only -0.5% at a depth of approximately 400 mbsf. From this depth toward the base of the core at 1000 mbsf, the isotopic composition increases to +0.5% (Fig. 4).

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Figure 1. Location of the sites drilled during Leg 133.

Site 815

This site exhibited relatively heavy O isotopic compositions (0.5%) to 1.2%) throughout the cored interval (Fig. 5 and Table 1).

Carbon Isotopic Data

Sites 811, 812, 813, and 814

With the exception of the upper portion of Sites 814 and 812, these sites all exhibit slightly negative (-1%) carbon isotopic compositions throughout (Fig. 6). Site 811 exhibits a slight enrichment down the core.

Sites 819, 820, 821, 822, and 823

All sites show a rapid depletion in \mathbb{S}^{13} C in the upper 10 to 50 mbsf, followed by an enrichment down the core. Site 819 exhibits the most severe depletion, attaining a δ^{13} C value of -15% at a depth of 25.48 mbsf. These depletions are less pronounced at other sites nearer to the Great Barrier Reef. In the lower portion of the core the δ^{13} C of the dissolved inorganic carbon (DIC) becomes isotopically heavier. The greatest enrichment occurs at Site 822, where values as heavy as +4‰ are reached. The magnitude of the enrichment in the lower portion of the core also changes with respect to the reef margin, being the highest at the sites farthest away from the margin (Fig. 7). In contrast to Site 822, Site 823 does not exhibit any enrichment in $\delta^{13}C$ down the core (Fig. 4 and Table 1).

Site 815

Site 815 exhibits a decrease in δ^{13} C throughout the upper 50 mbsf and maintains approximately the same isotopic composition to a depth of 350 mbsf. A slight enrichment is seen at the base of the core (Fig.5).

DISCUSSION

Carbonate Platform and Flank Sites 811/825, 812, 813, 814, 816, 817, and 818

The carbonate platform and flank sites can be separated into those situated on the Queensland Plateau and those located on the Marion Plateau. Sites 811, 812, 813, 814, 824, and 825 were cored on the Queensland Plateau and are all characterized by the present-day flux of substantial quantities of periplatform sediments (admixtures of bank-derived aragonite, HMC, and pelagic LMC). Sites 817 and 818



Figure 2. Oxygen isotopic data from pore waters recovered from Sites 811, 812, and 813.

were cored on the flanks of the Queensland Plateau. Sites 816 and 826 were drilled on the Marion Plateau.

Strontium

Previous ODP sites in the Bahamas (Leg 101) and the Maldives (Leg 115) where periplatform sediments have been cored (Swart and Guzikowski, 1988; Swart and Burns, 1990) were characterized by relatively high concentrations of Sr2+ in their interstitial pore fluids that resulted from the recrystallization of aragonite and HMC to LMC, and the consequent release of Sr²⁺ (Fig. 8). Therefore, perhaps the most surprising feature concerning the pore-water chemistry at Sites 811, 812, 813, and 814 is the absence of large increases in Sr²⁺ in the pore fluids, despite the fact that aragonite disappears relatively quickly with increasing depth sub-bottom (Fig. 9). We propose that fluid movement through the sediment prevents the establishment of normal diffusive Sr²⁺ gradients at these sites, an interpretation that can be supported by the following evidence: (1) the absence of geochemical gradients in the pore fluid; (2) the 87Sr/86Sr ratio of the pore water; and (3) the geothermal gradients. Perhaps the most persuasive evidence is the ⁸⁷Sr/86Sr ratio of the pore fluid. The pore fluids from Sites 811 to 814 all have 87Sr/86Sr ratios that are much more radiogenic than the ratio of contemporaneous sediments and probably reflect mixing between Sr released by recrystallization of carbonate sediments and Sr^{2+} contributed from modern seawater circulating through the platform (Elderfield, this volume) (Fig. 10). Such circulation does not appear to be taking place at a similar depth in the deeper flank sites, Sites 817 and 818, or at Site 816 situated on the top of the Marion Plateau. These sites exhibit steep diffusive Sr^{2+} gradients (Fig. 11B) and ${}^{87}Sr/{}^{86}Sr$ profiles (Elderfield et al., this volume) similar to those described for oceanic sediments (Gieskes et al., 1986) (Fig. 11A).

Oxygen and Carbon Isotopes

Although not conclusive, the oxygen and carbon isotopic profiles of these sites support the hypothesis of fluid movement in that there is little evidence for significant carbonate recrystallization. For example, if the pore fluids were being strongly influenced by carbonate recrystallization, they would rapidly become isotopically depleted in oxygen. This is not the case. Instead, the oxygen isotopic compositions of the pore fluids are relatively heavy throughout (Fig. 2). Carbon isotopic compositions are lighter than surface seawater but presumably similar to bottom water for the area (Fig. 6). If these fluids were being affected by carbonate recrystallization, they probably would show heavier carbon isotopic compositions. In fact, both Sites 812 and 814 exhibit some isotopically heavy values in the upper portion of the core, the same interval that exhibits Sr2+ enrichment. Site 811 displays a progressive enrichment down the core. These values suggest the presence of carbonate reactions in these regions, and, possibly, the presence of different aquifers. Different aquifers in the carbonate platform also are suggested by these oxygen isotopic data.

Calcium and Magnesium

Of further interest is the nature of the Ca²⁺, Mg²⁺, and Cl⁻ gradients. Sites 813 and 814 are characterized by steep Ca²⁺ gradients near the sediment/water interface, but relatively constant concentrations as depth increases sub-bottom. In contrast, Site 812 exhibits a constant concentration of Ca²⁺ as depth increases and a relatively sudden increase in Ca²⁺ concentration from 80 to 120 mbsf (Fig. 9). Normally, the geochemical gradients observed at Sites 813 and 814 are indicative either of fluid movement out of these sediments or of horizontal flow parallel to the scafloor. In contrast, the gradients at Site 812 are associated with the movement of normal seawater into the formation, either vertically from the overlying seawater or in a horizontal direction. Hence, these gradients support the interpretation of fluid flow through the Queensland Plateau sites. At Sites 816, 817, and 818, Ca²⁺ gradients do not support fluid movement and, therefore, concur with the Sr²⁺ data.

Geothermal Gradients

The interpretation of fluid flow at the Queensland Plateau sites is supported by geothermal measurements performed during the logging of Sites 812 and 814. Site 812 shows a relatively constant temperature of approximately 11°C to a depth of 200 mbsf and then a sudden rise to a temperature of approximately 17°C. At Site 814 there is a sharp temperature gradient close to the seawater/sediment interface. One problem with these data is that they were taken by logging immediately after drilling, and there is a considerable probability that the geothermal gradients are not representative of those normally present. Problems with these data can be seen when comparing Sites 817 and 812, which exhibit similar geothermal gradients in the open logged hole, yet have clearly different interstitial porewater profiles. Both of these sites probably penetrated under pressured zones, causing water to flow into the hole after logging. In the case of Site 812, this zone is probably close to the surface, thereby influencing water flow through the overlying veneer of uncemented sediments. At Site 817, however, this zone was considerably deeper;

Table 1. Summary of interstitial pore-wa	iter data from Leg 13.	3
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Sample	Depth	Age	Ph	Alk (mM)	Sal (g/kg)	CI [°] (mM)	Mg ²⁺ (mM)	Ca ²⁺ (mM)	SO ₄ 2- (mM)	K+ (mM)	Na ⁺ (mM)	Rb ²⁺ (μM)	Sr (µM)	C (‰)	0 (‰)	⁸⁷ Sr ^{/86} Sr	Error 1σ
825 Seawater	0.00	0.00	7.95	2.224	35.2	550	53.40	10.57	28.90				96	0.22			
825A-2W-1, 0-7	100.00	1.66	7.65	2.593	36.0	558	48.14	16.65	30.16				115	-0.34			
825A-4W-1, 0-7	150.00	2.49	7.63	2.634	35.8	556	48.35	16.63	30.27				118	0.37	11-2-5		
825A-5H-5, 145-150 825A-8H-5, 140-150	207.45	3.45	7.44	2.591	35.8	561	47.82	16.68	29.22				113	0.35	1.51		
825A-10X-2, 140–150	250.45	12.20	7.45	2.950	35.8	561	48.45	16.54	30.38				111	-1.69	1.60		
811A-1H-5, 145-150	4.45	0.40	7.57	3.106	36.0	543	49.80	13.80	28.80				149		0.12		
811A-2H-5, 145-150 811A-3H-5, 145-150	22.45	2.00	7.13	2 920	36.3	555	47.68	16.00	28.46				240				
811A-4H-5, 145-150	30.45	2.29	7.19	2.673	36.0	555	48.38	16.43	29.21				119		0.52		
811A-5H-5, 145-150 811A-6H-5, 145-150	39.95	2.64	7.14	2.547	36.0	557	48.40	16.42	28.86				114		0.49		
811A-7H-5, 145–150	60.45	3.39	7.09	2.489	36.0	556	48.72	16.26	28.92				117		0.45		
811A-8H-5, 145-150	69.95	3.45	7.19	2.549	36.0	551	49.07	16.70	29.71				114	-0.80	0.71		
811A-9H-5, 145-150 811A-10H-5, 145-150	79.45	4.00	7.13	2.549	36.0	557	48.82	16.56	29.51				115	-1.03 -0.99	0.94		
811A-11H-5, 145-150	98.45	5.60	7.12	2.605	36.0	557	48.66	16.24	29.50				119	-0.90	0.79		
811A-14H-5, 145-150	126.95	7.00	7.36	2.483	35.8	555	49.09	16.38	29.20				99	-0.23	0.57		
811A-20H-5, 145–150	183.95	9.00	7.40	2.505	36.0	542	48.18	16.66	29.72				112	-0.81	0.74		
811B-8V-5, 145-150	354.45	19.00	7.45	3.219	36.0	557	48.88	16.71						0.32	0.87	0 0001 50	0.000000
812A Surface seawater 812A-1H-2 145-150	0.00	0.00	7.96	2.543	25.8	551	54.01	10.36	29.62				91	-0.32	0.50	0.709168	0.000000
812C-1H-5, 145-150	4.45	0.08	7.56	2.904	35.0	543	53.38	10.34	29.02				102	-0.32	0.46	0.709103	0.000062
812A-2H-5, 145-150	12.35	0.69	7.42	3.212	35.8	545	53.13	10.43	29.32				114	-0.28	0.65	0.000070	
812C-2H-5, 145-150 812A-3H-5, 145-150	14.80	0.90	7.42	2.886	35.0	539	53.16	10.32	29.62				114		0.22	0.000078	
812C-3H-5, 145-150	24.37	1.51	7.31	2.871	35.0	542	53.01	10.46	29.85				114		0100	0.709083	0.000038
812C-5H-5, 145-150	36.97	2.31	7.43	2.701	35.2	542	52.85	10.32	29.62				102	0.39	0.32	0.709185	0.000028
812C-7H-5, 145–150	40.47	3.51	7.27	2.834	35.0	543	52.79	10.62	29.00				102	-0.37	0.05	0.709188	0.000028
812C-8H-5, 145-150	64.47	4.05	7.35	2.902	35.2	543	52.62	10.79	29.19				102	-0.78	0.84	0.709173	0.000052
812C-9H-5, 145-150 812C-10H-5, 145-150	74.97	4.71	7.34	2.863	35.2	534	52.86	10.62	29.66				102	-0.27	0.57	0.709136	0.000078
812A-12X-5, 145-150	91.05	5.73	7.27	3.107	35.8	545	52.41	10.84	30.02				112	0.55	0.54	0.709159	0.000060
812C-13H-5, 145-150	112.97	7.12	7.45	2.910	35.2	541	52.38	11.42	29.03				102		0.86		
813A Seawater 813A-1H-3 145-150	0.00	0.00	7.99	2.455	36.2	545	54.46	10.46	29.89				96 134	-0.21		0 709149	0.000042
813A-2H-5, 145-150	13.17	0.45	7.34	2.927	36.0	548	52.66	11.45	29.24				111	0.21	0.20	0.107117	0.000012
813A-3H-5, 145-150	22.67	0.93	7.17	2.501	35.8	543	51.95	11.30	30.55				107		0.41	0.709147	0.000024
813A-5H-5, 145–150 813A-5H-5, 145–150	41.67	1.27	7.16	2.506	35.2	540 546	51.31	11.22	28.04				104		0.82	0.709147	0.000020
813A-6H-5, 145-150	51.17	2.42	7.07	3.199	35.8	542	51.52	11.09	29.04				104	-1.17	0.33		
813A-7H-5, 145-150 813A-8H-5, 145-150	60.67	2.50	7.28	2.187	35.2	542	51.86	11.15	29.83				105	-1.09	0.10	0.709137	0.000032
813A-9H-5, 145–150	79.67	3.00	7.31	2.584	35.2	540	51.90	11.18	29.05				105	-1.06	0.08	0.709194	0.000038
813A-10H-5, 145-150	89.17	3.51	7.21	2.557	35.2	543	51.73	11.27	29.84				110	-0.63	0.02	0.709138	0.000020
813A-12H-5, 145–150 813A-15H-5, 145–150	136.67	7.00	7.34	2.767	35.2	538	51.08	12.10	29.91				99	-1.00	0.24	0.709124	0.000050
813A-18H-5, 145-150	165.17	7.50	7.36	2.608	35.2	543	51.02	11.99	29.76				99	-1.22	0.30		
813A-21H-5, 145-150 814 Seawater	193.67	9.50	7.62	2.755	35.2	544	51.15	11.86	31.20				99 96	-0.66	0.23	0 709163	0.000020
814A-1H-3, 145-150	4.45	0.28	7.47	3.900	35.5	519	53.62	10.34	29.08				167	-0.22		0.709173	0.000020
814A-2H-5, 145-150	13.37	0.47	7.31	3.260	35.0	544	53.42	10.78	28.37				181	0.35	-0.17	0.709165	0.000020
814A-3H-5, 145–150 814A-4H-5, 145–150	32.37	1.88	7.20	3.196	35.2	539	53.83	11.02	29.25				167	0.54		0.709104	0.000028
814A-5H-5, 145-150	41.00	2.42	7.16	3.123	35.5	546	53.14	11.83	28.70				125	10000		0.709140	0.000024
814A-6H-5, 145-150 814A-7H-2, 145-150	51.37	2.60	7.14	2.882	35.5	560	52.95	11.88	29.29				108	-0.28		0.709135	0.000020
814A-9H-5, 145-150	73.97	6.27	7.52	3.139	35.2	538	53.09	11.51	28.59				95			0.709150	0.000024
814A-10H-5, 145-150	83.47	7.76	7.52	3.201	35.5	559	52.84	11.60	29.74				101	-0.48		0.709131	0.000028
814A-13H-5, 145-150 814A-23X-1, 145-150	200.10	12.20	7.04	2.557	35.5	549	53.05	11.87	28.25				97	-0.65		0.709121	0.000018
814A-24X-2, 145-150	209.80	16.26	7.37	2.776	35.2	572	52.15	12.06	28.46				93	-1.21		0.709134	0.000028
814A-26X-3, 145-150 814A-28X-2, 145-150	230.20	17.10	7.27	2.608	34.0	552	52.34	11.87	28.25				92	-0.44		0.709139	0.000020
814A-29X-2, 145-150	257.20	18.22	7.38	3.029	35.5	546	52.71	11.94	26.97				93	-1.00		0.709155	0.000020
815A Surface seawater	0.00	0.00	8.24	2.646	35.0	573	53.55	10.27	28.12	9.80		1.4	86	0.11			
815A-2H-5, 145-150 815A-3H-5, 145-150	14.25	0.47	7.28	3.082	35.5	537	48.59	10.28	24.16	10.84		1.4	335	-0.81	1.12		
815A-4H-5, 145-150	33.25	2.42	7.32	3.414	35.8	573	44.56	10.15	20.09	9.82		1.4	526	-2.17	0.76	1502020000	1212225223
815A-5H-5, 145-150	42.75	2.50	7.32	3.517	35.8	583	40.90	10.14	18.36	9.52		1.3	610	-2.34	1.23	0.708989	0.000021
815A-7H-5, 145–150	61.75	2.55	7.23	4.046	36.0	592	35.37	10.45	14.75	8.75		1.1	817	-1.88	0.02		
815A-8H-5, 145-150	71.25	2.64	7.25	4.162	35.8	607	34.13	10.81	12.97	8.05		1.0	897	-2.08	0.85		
815A-9H-5, 145-150 815A-10H-5, 140-150	80.75	2.67	7.32	4.408	36.0	613	32.37	10.75	0 73	7.71		1.0	973	-2.33	0.89	0.709001	0.000038
815A-13H-5, 140-150	118.70	2.81	7.25	4.686	36.0	603	27.23	10.38	5.76	6.87		0.8	1239	-1.98	0.66	0.709001	0.00030
815A-16H-5, 140-150	147.20	2.92	7.46	4.511	36.5	638	25.31	10.59	2.76	5.69		0.7	1347	1.0	0.72	0 700007	0.000001
815A-19H-5, 140-150 815A-22H-5, 140-150	204.20	3.13	7.35	4.318	37.8	698	25.44	11.16	1.63	5.59		0.6	1445	-1.62 -1.81	0.69	0.708907	0.000021
815A-26X-3, 140-150	226.90	3.21	7.46	4.144	40.0	705	28.68	14.19	2.33	4.33		0.5	1541	-2.18	0.59	0.708897	0.000018
813A-28X-3, 140-150	246.20	3.28	1.56	4.005	40.0	699	30.54	15.24	4.11	4.62		0.5	1412	-2.57			

Table 1 (continued).

Sample	Depth	Age	Ph	Alk (mM)	Sal (g/kg)	Cl' (mM)	Mg ²⁺ (mM)	Ca ²⁺ (mM)	SO ₄ 2- (mM)	K+ (mM)	Na ⁺ (mM)	Rb ²⁺ (µM)	Sr (µM)	C (%)	0 (‰)	⁸⁷ Sr ^{/86} Sr	Error 1σ
815A-31X-4, 140–150 815A-34X-5, 140–150 815A-37X-5, 140–150	276.70 307.30 336.40	3.39 3.51 4.60	7.38 7.46 7.48	3.218 2.575 2.451	42.0 42.2 43.2	714 742 744	31.96 35.49 37.80	20.19 20.72 22.14	5.69 7.81 9.97	4.67 5.26 5.85		0.5 0.5 0.6	1400 1414 1261	-2.14 -2.36 -2.55	0.53 0.69	0.708908	0.000021
815A-40X-5, 140–150 815A-43X-5, 140–150 815A-47X-5, 140–150 816 Sequence	365.50 394.50 432.70	5.26 5.90 6.74	7.17 7.23 7.26	3.226 3.105 3.107	44.2 46.5 45.5	760 770 760	37.81 40.06 42.61	24.27 26.06 26.82	11.38 13.89 15.99	5.25 5.72 6.43		0.9 0.8 1.1	1342 1124 920	-3.61 -2.39 -1.11	0.50 0.44	0.708864	0.000021
816A-1H-3, 145–150 816A-2H-5, 145–150 816A-3H-5, 145–150	4.45 11.95 22.45	0.93 1.88 2.60	7.47 7.22 7.30	3.381 3.000 3.077	35.8 35.8 36.0	538 548 559	52.54 51.64 51.00	10.73 11.58 12.34	28.47 29.60 29.56				149 210 240	-0.55 -0.82 -1.70	0.32 0.35 1.40		
816A-4H-5, 145–150 816A-5H-5, 145–150 816A-6H-5, 145–150	31.95 41.45 50.95	3.50	7.28 7.34 7.24	3.306 3.153 3.240	36.4 37.0 37.5	562 563 571	50.60 49.43 49.06	13.09 14.15 14.98	29.46 29.45 29.01				297 328 336	-2.22 -1.44 -1.57	0.94 1.04 0.55		
816A-7H-2, 145–150 816A-8H-5, 145–150 816A-9H-5, 145–150 816A-10H-5, 145–150	60.45 69.95 79.45 88.95		7.14 7.25 7.26 7.39	3.260 3.058 3.038 3.173	38.0 37.8 37.0 36.5	572 584 572 562	48.62 48.42 47.62 49.18	15.90 16.65 17.00	28.82 28.39 27.95 29.31				348 349 318 223	-1.51	0.54 0.65		
817A-1H-3, 145–150 817A-2H-5, 145–150 817A-2H-5, 145–150 817A-3H-5, 145–150	4.45 13.15 22.65	0.28 0.34 0.47	7.65 7.45 7.47	3.365 4.966 7.137	35.0 35.2 35.2	551 557 559	52.63 55.10 55.56	9.33 7.81 7.28	29.18 29.25 28.38				127 201 350	-1.02	0.08		
817A-4H-5, 145–150 817A-5H-5, 145–150 817A-6H-5, 145–150	32.15 41.65 51.15	0.93 1.04 1.16	7.53 7.54 7.32	8.013 8.099 7.898	35.5 36.0 35.5	562 563 559	55.70 55.03 54.49	7.66 8.17 8.19	28.05 28.81 29.05				466 544 517		0.45	0.709138 0.709145	0.000024 0.000026
817A-7H-1, 0–7 817A-7H-5, 145–150 817A-8H-5, 145–150 817A-9H-5, 145–150	53.20 60.65 70.25 79.65	1.18 1.27 1.66	7.67 7.43 7.37	7.868 7.390 6.771	34.5 35.5 35.0	558 561 559	54.49 54.57 53.89	8.33 8.25 8.39	28.42 29.56 28.57 20.16				531 530 512		0.77	0.709118	0.000036
817A-10H-5, 145–150 817A-11H-5, 145–150 817A-12H-1, 0–7	89.15 98.65 100.70	2.42 2.59 2.63	7.34 7.61 7.74	6.127 5.909 6.127	35.2 35.0 36.2	562 559 554	54.01 53.80 53.32	9.02 9.08 10.03	28.78 29.20 29.20				475 468 478		0.14 0.25	0.709052	0.000024
817A-14H-5, 145–150 817A-16H-5, 145–150 817A-17H-1, 0–7	127.15 146.15 148.20	3.11 3.45 3.88	7.39 7.35 7.43	5.575 5.740 6.065	36.0 35.2 35.8	562 553 555	52.11 51.59 51.52	11.08 11.40 12.00	29.91 30.06 29.20				473 446 472		0.32	0.709016	0.000020
817A-19H-5, 145–150 817A-21H-5, 140–150 817A-22H-1, 0–7	174.62 193.60 195.70	4.60 6.74 6.97	7.50 7.17 7.62	5.443 5.821 4.600	35.5 35.2 35.5	558 559 558	50.61 49.27 50.06	13.26 14.71 14.72	30.13 30.28 31.06				432 403 393		0.19	0.708965 0.708916	0.000020 0.000020
817A-24X-5, 140–150 817A-26X-5, 140–150 817A-27X-1, 0–7	241.40 243.70	10.48 13.00 13.36	7.29 7.39 7.58	4.729 4.376 3.998	35.5 36.0 36.2	563 564	49.31 48.39 50.02	15.72 16.41 15.24	29.98 30.55 30.90				367 314		0.03	0.708833	0.000018
817A-29X-5, 140–150 817A-5R-3, 140–150 818B Seawater	270.40 303.40 0.00	16.92 17.00 0.00	7.42 7.26 8.20	4.043 3.908 2.595	37.0 37.5 35.5	570 575 548	48.50 46.51 54.78	16.76 19.27 10.64	30.76 30.00 29.03				315 365 97	0.00	0.44	0.708824 0.708826 0.709173	0.000020 0.000030 0.000022
818B-1H-4, 145–150 818B-2H-5, 145–150 818B-3H-5, 145–150 818B-3H-5, 145–150	4.49 12.99 22.46	0.28	7.46 7.59 7.30	3.488 3.232 4.699	34.8 35.0 36.2	535 537 540	52.94 53.61 54.82	10.03 9.99 8.79	27.33 27.98 27.72				196 211 449	0.29 0.94		0.709118	0.000040
818B-5H-5, 145-150 818B-6H-5, 145-150 818B-7H-5, 145-150	41.49 50.99 60.49	0.93	7.30	6.373 6.289 6.519	36.0	555 555 562	54.30 53.77 53.70	8.61 9.12 9.97	26.57 26.42 26.76				662 685 704	0.81		0.709114	0.000030
818B-8H-5, 145–150 818B-9H-5, 145–150 818B-10H-5, 145–150	69.99 79.49	1.48	7.72	6.679	36.2	562 564 572	61.97 52.45	11.04 11.24	26.93 26.66 26.56				716 685 746	0.39		0 709108	0.000028
818B-10H-5, 145-150 818B-13H-5, 145-150 818B-16H-5, 145-150 818B-19H-5, 145-150	120.53 148.91 177.50	2.20 2.40 2.60	7.27	6.471 6.269 5.787	38.0 38.2 38.5	572 587 606 615	52.39 52.50 53.34	12.73 13.44 14.77	26.60 27.59 28.25				716 704 638	0.24		0.709108 0.709036 0.709148 0.709048	0.000024 0.000064 0.000020
818B-22H-5, 145–150 818B-25H-5, 145–150 818B-28H-5, 145–150	206.00 234.51 262.00	3.00 3.20 3.51	7.30 7.24 7.29	5.057 4.412 3.413	40.2 40.2	621 634 620	53.53 53.46 51.68	15.75 18.32 21.29	28.54 29.83 32.21				601 599 470	-0.77		0.709055 0.709032 0.709014	0.000018 0.000036 0.000018
818B-31H-5, 145–150 819 Seawater 819A-1H-5, 145–150	291.00 0.00 5.95	3.88 0.00 0.28	7.30 8.27 7.21	2.862 2.655 9.584	35.5 34.2	606 551 551	50.25 53.68 48.54	20.94 10.36 6.26	31.21 29.74 18.11	10.62 12.30	474 474		217 98 80	-0.55 1.87 -12.32	0.48	0.708971	0.000021
819A-2H-5, 145-150 819A-3H-5, 145-150 819A-4H-5, 145-150	15.95 25.45 34.95	0.32 0.36 0.39	7.31 7.43 7.33	10.926 11.059 9.500	32.5 31.8 31.8	550 548 550	39.39 30.63 25.42	4.16 3.32 4.95	6.97 0.00 0.00	10.08 10.55 9.33	476 478 487		95 112 179	-0.93 -15.00 -5.03	1.18		
819A-5H-5, 140–150 819A-6H-5, 140–150	44.45 53.95	0.43	7.53	8.388 7.652	31.6 30.5	546 542	23.32 19.79	4.60	0.00	9.02 8.22	486 490		234 273	-1.83	0.66		
819A-8H-5, 140–150 819A-8H-5, 140–150 819A-9H-5, 140–150	72.90 82,40	0.51 0.55 0.58	7.68 7.53	7.936 8.341	30.5 31.0 31.0	548 535 541	16.83 16.05	3.45 3.41	0.00	7.18 7.02	491 499		242 221 241	0.06 0.05	0.00 0.28 0.12		
819A-10H-5, 140–150 819A-13H-5, 140–150 819A-16H-6, 140–150	91.90 118.40 137.80	0.62	7.84	8.985 9.988	31.0 30.2	539 548	15.26 13.89	2.86 2.97	0.00	6.79 5.64	501 514		231 292	-0.51	0.35		
819A-19X-2, 140–150 819A-23X-5, 140–150	168.80 205.50	0.92 0.93	7.58 7.71	11.879 12.423	30.5 31.0	539 537	17.68 18.61	3.86 4.42	2.00 2.47 1.27	5.41 4.82	502 497		483 260	0.26	-0.08 -0.19		
819A-25X-1, 140–150 819A-28X-1, 140–150	224.40 243.30	0.93	7.51	11.559	31.0 31.0	544 541	19.27 25.37	4.81	1.95 6.23	4.99	502 491		252 171	0.22	-0.29 -0.45		
819A-31A-2, 140-150 819A-34X-2, 140-150 819A-37X-5, 140-150	301.30	1.02	7.58	9.691	31.0 31.0	535 534	18.31 22.07	4.35	3.06	3.62	493		155 277 174	0.28	-0.31		
819A-40X-6, 140–150 819A-43X-5, 140–150	358.90 387.80	1.15	7.68	7.819	31.5	544 550 541	22.08	5.83	7.13	3.34	502		176	1.18	0.00		
820 Seawater 820A-1H-5, 145-150	0.00	0.00	8.10 7.47	2.574	35.0	556	52.50 53.80	10.56	29.39 24.91	10.68	480		114	1.20	0.47		
820A-2H-5, 145-150 820A-3H-5, 145-150	14.65 24.15	0.12 0.20	7.56 7.52	5.386 5.501	36.2 34.0	560 562	51.13 43.31	6.96 6.88	22.46 16.62	10.11 9.41	484 491		126 175	-7.83 -0.79	0.27		

Table 1 (continued).

Sample	Depth	Age	Ph	Alk (mM)	Sal (g/kg)	Cl ⁻ (mM)	Mg ²⁺ (mM)	Ca ²⁺ (mM)	SO ₄ 2- (mM)	K+ (mM)	Na ⁺ (mM)	Rb ²⁺ (µM)	Sr (µM)	C (‰)	0 (‰)	⁸⁷ Sr ^{/86} Sr	Error 1σ
820A-4H-5, 145-150	33.65	0.28	7.48	6.157	33.0	558	34.33	6.22	9.63	7.98	494		210	-1.69	0.55		
820A-5H-5, 145-150 820A-6H-5, 145-150	43.15	0.31	7.41	6.490	32.0	554	27.07	5.20	3.46	7.51	494		241	-1.16 -1.42	0.53		
820A-7H-5, 140–150	62.10	0.37	7.94	7.901	31.8	555	20.69	4.62	0.80	6.36	507		227	-2.39	0.00		
820A-8H-5, 140-150 820A-9H 5, 140-150	71.60	0.41	7.63	7.700	31.5	544	18.91	4.05	0.00	6.31	499		271	-1.10	0.43		
820A-10H-5, 140–150	90.60	0.47	7.67	6.365	31.2	543	12.80	3.15	0.00	5.58	511		304	-0.44			
820B-12H-1, 0-7	103.20	0.53	7.79	6.652	31.0	540	14.80	3.53	3.23	5.50	510		331	-1.04	0.44		
820A-16V-1, 140–150	141.60	0.60	8.02	5.613	30.2	542	9.58	3.01	0.88	4.87	520		426		0.14		
820B-17X-1, 0-7	150.70	0.74	7.95	3.217	34.0	547	43.47	8.24	2.00	8.42	442		251		0.02		
820B-17X-3, 140–150 820B-20X-5, 140–150	155.10	0.75	7.88	0.726	31.0	537	7.25	2.44	0.29	2.89	520 525		400	-0.71	-0.19		
820B-23X-1, 0-7	203.40	0.97	8.44	7.480	32.5	539	15.85	4.20	7.93	4.29	517		369	0.00	0.25		
820B-23X-2, 140-150 820B-27X-3, 140-150	206.30	0.98	7.89	6.981	30.2	538 540	4.88	2.31	0.43	2.65	528		416	-0.92	0.18		
820B-28X-1, 0-7	251.10	1.18	8.16	0.217	31.0	551	16.01	4.74	9.90	3.25	525		473				
820B-30X-1, 140-150 820B-33X-1 0-7	271.80	1.27	7.94	5.382	31.8	550	11.20	4.32	5.86	3.08	532 476		421		0.54		
820B-33X-1, 140-150	300.80	1.32	8.23	6.604	30.0	535	5.77	2.81	0.00	2.03	521		554				
820B-36X-5, 140-150	335.90	1.39	8.14	5.716	30.5	542	9.05	2.94	2.62	2.43	525		504	0.77	0.97		
820B-42X-4, 140-150	382.60	1.44	7.89	6.195	31.0	536	9.41	3.63	2.20	2.38	517		575	-0.77	0.69		
821 Seawater	0.00	0.00	8.12	2.595	35.2	542	53.51	10.53	30.15	10.66	466		93	1.73	0.76		
821A-2H-5, 145-150	2.95	0.02	7.40	0.996	35.8	561	54.65	6.25	20.19	9.04	487		86	-0.60	0.75		
821A-3H-5, 145-150	21.35	0.18	7.32	4.817	35.2	558	49.38	7.80	24.47	9.79	487		147	-0.69	1.10		
821A-4H-5, 145–150 821A-5H-5, 145–150	30.85	0.28	7.33	5.086	35.2	560	· 44.40 35.44	8.05	21.73	9.04	494		329	-4.93 -0.33	1.71		
821A-6H-5, 145-150	49.85	0.33	7.45	5.267	33.8	551	29.74	6.14	15.65	6.08	509		353	0.89	1.95		
821A-7H-5, 140–150 821A-8H-5, 145–150	59.30 68.85	0.37	7.29	6.047	33.5	551 545	24.85	5.96	13.29	6.38 5.76	515 514		379	-0.70	2.04		
821A-9H-5, 145–150	78.35	0.43	7.40	5.382	33.0	548	18.46	4.91	11.17	5.15	523		437	0.24	0.39		
821A-10H-5, 145-150 821A-13H-5, 145-150	87.85	0.46	7.68	5.887	32.0	544	15.58	4.43	9.22	5.21	522		436	-0.36	0.38		
821A-17X-5, 140–150	144.80	0.71	7.89	6.330	30.2	538	5.08	2.32	0.00	2.68	526		462	0.02	0.91		
821A-20X-5, 140-150	182.30	0.93	7.82	6.322	30.2	539	6.96	2.84	0.31	3.18	522		548	0.21	0.33		
821A-26X-5, 140–150 821A-26X-5, 140–150	239.60	1.13	8.11	5.745	30.5	538	5.05	2.84	0.00	1.79	527		759	-0.64	0.63		
821A-29X-5, 140-150	268.50	1.26	8.20	4.360	30.2	540	4.74	2.66	0.11	1.67	527		702	-0.48	0.50		
821A-33X-5, 140–150 821A-35X-3, 140–150	307.10	1.33	8.23	5.266	32.2	536 538	7.56	3.29	1.60	1.82	520 522		612 695	-0.70	0.46		
821A-38X-5, 140-150	355.30	1.42	8.10	5.012	31.5	535	8.71	3.79	0.83	1.61	514		642	-0.97	0.39		
821A-41X-4, 140–150 822 Seawater	382.70	1.48	8.33	3.650	30.2	538 546	6.87 54 61	4.03	1.33	1.41	517		730	-1.35	0.67		
822A-2H-5, 145-150	8.40	0.14	7.58	9.862	33.0	538	42.94	4.28	10.54	11.43			83	0.00	0.15		
822A-3H-5, 140-150 822A-4H-5, 140-150	17.80	0.28	7.55	13.828	32.5	547 544	34.64	4.27	0.00	10.04			123	-1.82 -6.35	0.33		
822A-5H-5, 140–150	36.80	0.55	7.32	11.822	32.0	545	30.31	4.35	0.00	8.95			161	-0.55	0.25		
822A-6H-5, 140–150 822A-7H-5, 140–150	46.30	0.62	7.52	10.469	32.0	535	29.26	4.19	0.00	7.83			154	-5.85	0.09		
822A-8H-5, 140–150	65.30	0.78	7.45	8.561	31.0	533	26.22	4.48	0.00	7.77			140		0.11		
822A-9H-5, 140-150	74.80	0.85	7.72	9.270	31.8	543	26.83	3.96	0.00	8.08			138	0.26	0.77		
822A-11H-5, 140–150	93.80	0.95	7.62	9.498	31.5	543	26.05	3.95	0.00	6.40			137	1.16	0.95		
822A-12H-1, 0-7	95.90	0.95	7.70	3.571	35.0	563	49.48	9.39	26.88	5.60			166		0.39		
822A-14X-4, 140–150 822A-17X-5, 140–150	122.80	1.01	7.68	10.575	32.0	536 540	24.72	4.62	1.24	5.41			136	3.27	0.01		
822A-18X-1, 0-7	154.00	1.08	8.12	5.981	34.0	506	22.51	4 40	0.00	5.04			126		0.97		
822A-23X-1, 0-7	192.20	1.13	8.70	8.093	35.0	530	23.51	4.49	0.00	5.06			120		-0.87		
822A-23X-2, 140-150	195.10	1.16	7.69	8.347	31.8	544	23.72	4.84	0.00	4.64			132	0.00	-0.72		
822A-27A-5, 140–150 822A-28X-1, 0–7	247.60 249.80	1.27	7.86	8.571 6.904	31.2	550 545	23.18	5.32	1.02	4.44			118	2.55	-0.72		
822A-29X-5, 140-150	266.90	1.31	7.83	7.049	32.2	562	23.27	4.85	2.36	4.64			116	2.95	-0.70		
822A-32X-4, 140–150 822A-33X-1, 0–7	294.30 298.10	1.37	7.82	8.388	31.5	545 542	22.74	5.94	1.12	3.38			158		-0.95		
822A-35X-4, 140-150	323.40	1.74	7.70	5.941	31.5	548	21.35	5.40	0.64	3.32			214		-0.89		
822A-38X-1, 0-7 822A-38X-4 140-150	346.30	1.98	8.27	4.178	34.2	545	40.09	9.29	21.47	5.20			187				
822A-41X-3, 140–150	379.50	2.33	7.69	5.769	31.8	551	21.22	5.65	3.45	3.14			290		-0.97		
822A-44X-4, 140-150	405.90	2.60	7.85	5.050	31.8	547	18.72	6.54	0.34	2.62			388		-0.91		
823 Seawater	0.00	0.00	7.93	2.456	35.0	545	53.56	10.47	30.22	10.31	469		105		-0.94	0.709160	0.000020
823A-1H-3, 145-150	4.45	0.14	7.51	4.421	35.0	543	52.11	9.31	26.80	11.42	466		133	-1.94	0.38	0.709165	0.000023
823A-3H-5, 145–150	22.75	0.21	7.40	4.035	33.8	540	47.22	5.32	15.72	12.07	468		148	-2.24	0.89		
823A-4H-5, 145-150	32.25	0.47	7.50	9.881	32.5	543	39.00	3.13	6.93	11.43	469		190	-3.10	0.56		
823A-5H-5, 145-150 823A-6H-5, 145-150	41.75 51.25	0.57	7.43	12.196	31.8	550	33.03	3.72	0.67	9.74	479		202	-1.24	0.10	0.709155	0.000018
823B-7H-1, 0-7	55.80	0.71	8.01	12.550	31.8	545	31.55	4.12	0.00	8.24	475		241	-0.92			
823A-7H-5, 145-150 823A-8H-5, 145-150	60.75 70.25	0.76	7.39	11.421	31.8	544	29.67	4.36	0.00	9.84	475		220	-2.39	0.06		
823A-9H-5, 145-150	79.75	0.93	7.52	9.538	32.0	547	28.18	3.52	0.00	10.05	480		201	-1.60	-0.14		
823A-10H-5, 145-150	89.25	1.03	7.78	8.922	31.8	547	27.61	3.39	0.00	9.95	480		196	-1.55			

Sample	Depth	Age	Ph	Alk (mM)	Sal (g/kg)	Cl [°] (mM)	Mg ²⁺ (mM)	Ca ²⁺ (mM)	SO ₄ 2- (mM)	K+ (mM)	Na ⁺ (mM)	Rb ²⁺ (μM)	Sr (µM)	C (‰)	0 (‰)	⁸⁷ Sr ^{/86} Sr	Error 1σ
823B-12H-1, 0-7	103.30	1.17	8 33	9 912	31.8	542	28 42	4.83	0.43	5 50	478		212				
823B-12H-5, 140-150	110.70	1.24	7.82	7 745	32.2	542	25 91	3.97	0.00	8 11	478		197	-1.85	0.13		
823A-13H-5 145-150	117.95	1.27	7.87	7 929	31.8	545	25.18	3.07	0.00	8 74	483		216	-1.57	0.06		
823B-15X-5 140-150	139 50	1.36	7.82	7 668	31.8	547	26.15	3.77	0.00	7.25	484		234	-0.41	0.04	0 709062	0.000024
823B-17X-1 0-7	151.40	1 48	7 70	4.450	34.0	548	44 72	7 73	20.38	7.56	477		211	0.61	0.04	0.707002	0.000024
823B-18X-3 140-150	165 50	2 29	7.66	7.651	31.5	553	25.08	4.05	0.65	6 44	401		282	0.01			
823B-21X-3 140-150	194 60	2 34	7.92	6.874	32.0	540	23.74	4.14	0.00	8 21	488		374	0.72	-0.37		
823B-22X-1 0-7	199.80	2 38	7 59	7 914	32.0	547	25.94	4.14	0.50	5 58	485		409	-0.72	0.57		
823B-24X-2 140-150	221 60	2.42	7.76	6.081	31.8	547	20.96	5 14	0.00	5.25	401		477	-1.16	-0.60	0 708990	0.000026
823B-26X-1 0-7	237.70	2.60	8.03	6.822	33.8	543	27 50	5 20	4 32	4 31	485		505	1.10	0100	01100330	010000000
823B-27X-6 140-150	256 20	2.65	7.86	5 968	31.8	548	21 41	3.81	0.00	4.73	404		432	-1 14			
823B-30X-2 140-150	279 20	2.87	7.82	4 059	31.8	552	21.91	3.80	0.35	4 30	407		619	1.14	-0.30	0 708962	0.000018
823B-32X-1 0-7	295 60	3.03	8.03	5 752	31.8	543	23.73	5.06	3.60	3.66	480		662		0.50	0.700702	0.000010
823B-33X-4 140-150	311 20	3 18	7.91	3 780	32.0	546	20.44	5.21	2.42	3.08	495		687	-2 52	-0.35		
823B-36X-5, 140-150	341.60	3.48	7.91	3.068	31.8	558	18 45	4.82	0.04	4 43	506		703	-0.85	-0.33		
823B-39X-5, 140-150	370.40	3 51	7.97	3 537	31.8	554	10.45	4 64	0.00	3.49	502		654	-1.18	-0.51	0 708878	0.000026
823B-42X-4 140-150	397.90	3 70	7 84	3 803	31.8	551	18 67	5 73	0.45	2.95	400		742	-1.18	-0.50	0 708846	0.000040
823B-45X-5 140-150	428 30	3.88	7 73	3 528	31.8	540	17.23	5 67	0.00	2.84	408		825	-3.03	-0.58	0 708840	0.000030
823B-48X-4, 140-150	455.80	3.95	7 70	2 950	31.8	556	15.81	6.16	0.19	2.60	508		029	0.56	-0.11	0.700010	01000000
823B-52X-4 140-150	493 70	415	7.83	2 838	31.8	560	13.26	7.41	0.00	2.45	514	1	185	-1.59	-0.52	0 708855	0.000030
823B-55X-5, 140-150	525.20	4.24	1.05	2.000	32.0	557	12.99	7.91	0.00	1 70	508		269	-1.64	-0.50	0.708878	0.000026
823B-58X-5, 140-150	552.80	4.40	7 87	2 4 3 1	32.0	560	13.12	8 43	0.48	1.69	513	1	338	1.01	0.04	0 708904	0.000024
823B-61X-5 135-150	581.65	4 50	7 72	2 335	32.2	500	1.5.14	0.45	0.40	1.07	515		550	-1.43	0.01	0 708879	0.000028
823B-64X-1 135-150	604 55	4 60	7 91	2 081	32.2	563	13 71	0.00	1.08	2 27	513	1	515	-1.51	-0.48	0 708850	0.000022
823B-67X-5_135-150	639 55	5.06	7.90	1.836	32.2	571	14 18	9.20	0.69	2 25	519	1	780	1.51	-0.42	0.1000000	0.000022
823B-70X-5 110-125	667.70	5.26	7.81	1 807	32.2	568	13.42	10.27	1.20	2 33	515		870		-0.37		
823B-73X-2 135-150	692.45	5 40	7.87	1 910	32.2	572	14 40	10.27	1.14	2 44	515		013		-0.15	0 708884	0.000034
823B-76X-4, 135-150	724.45	5.60	7.79	2.848	32.2	568	14.02	11.91	0.55	2.28	510		248	-3.86	0110	0.708879	0.000022
823B-79X-4, 135-150	753.45	5.90	7.70	2.680	32.4	572	14.54	13.42	1.37	2 65	511		335	5.00	-0.17	0.100012	01000044
823B-82X-1, 135-150	778.05	6.31	7.56	3.181	32.5	568	16.60	12.24	1.96	2.13	507		2410		-0.11		
823C-1R-3, 130-150	788.30	6.48	7.91	2.258	32.5	569	17.15	12.25	2.60	3.10	507		2162	-2.22	-0.07		
823C-4R-1, 130-150	913.90	8.54		ana o	33.0	571	17.65	13.63	2.79	3.73	505		2023				
823C-7R-2, 130-150	844.50	7.40			33.8	566	18.95	14.70	2.92	3.75	495		2248		0.00		
823C-10R-1, 125-150	872.05	7.85			33.8	576	17.05	16.17	1.92	2.74	504		2444		0.16		
823C-13R-4, 130-150	905.50	8.40			35.0	576	16.60	17.70	1.78	2.34	501		2785		0.14		
823C-16R-4, 130-150	934.60	8.88			35.0	570	18.57	18.11	2.84	2.16	494		2774				
823C-19R-3, 130-150	961.60	9.33			34.0	577	20.33	19.77	3.97	2.88	495		2750		0.36	0.708819	0.000018
823C-22R-3, 130-150	990.50	11.60			33.8	570	21.88	18.00	5.63	3.68	491		2877		1.47	0.708804	0.000018
824 Seawater	0.00	0.00	7.99	2.745	35.2	556	54.87	10.44	29.41	10.30			110			0.709160	0.000020
824B-1H-2, 145-150	2.95	0.03	7.62	3.650	35.0	541	54.12	10.30	29.19	12.17			203	2.24	0.20	0.709142	0.000026
824B-2H-5, 145-150	12.45	0.12	7.37	3.406	35.5	541	54.72	10.06	29.62	10.81			323	0.78	0.19	0.709172	0.000026
824B-3H-5, 145-150	21.95	0.20	7.40	3.388	35.2	542	54.16	10.30	29.67	10.43			326		0.27	0.709192	0.000026
824B-4H-5, 145-150	31.45	0.29	7.60	2.994	35.2	542	54.11	10.26	29.73	10.17			249		0.24	0.709124	0.000036
824B-5H-5, 145-150	40.95	0.38	7.58	3.066	35.2	541	54.29	10.37	29.20	11.68			285			0.709107	0.000034
824B-6H-5, 145-150	50.45	0.47	7.41	3.064	35.2	539	53.97	10.45	29.00	10.35			208	0.84	0.10	0.709100	0.000032
824A-1H-5, 140-150	57.40	0.63	7.57	3.339	35.0	548	54.19	10.47	29.52	11.48			277	-0.60	0.20	0.709120	0.000022
824A-2H-5, 140-150	66.90	0.84	7.35	3.037	35.0	546	54.81	10.50	28.97	10.71			271			0.709108	0.000022
824A-3H-4, 140-150	74.90	1.02	7.29	3.375	35.0	541	54.29	10.46	30.00	9.85			270			0.709127	0.000044
824A-5H-5, 140-150	95.40	1.48	7.69	2.906	35.0	541	53.23	10.49	29.89	10.11			265	1.24		0.709111	0.000022
824A-8H-2, 140-150	119.40	2.00	7.62	3.100	35.0	546	53.83	10.67	29.38	10.26			280	1.22		0.709116	0.000018
824A-14H-5, 140-150	173.10	5.26	7.52	2.973	35.0	551	54.45	10.74	29.63	11.07			223		0.19	0.709112	0.000024
824A-17X-2, 145-150	195.50	6.00	7.63	3.106	35.8	556	55.58	10.74	30.47	11.89			212	0.24			
824A-21X-2, 140-150	235.60	8.00	7.65	3.062	35.8	560	55.68	10.79	29.96	11.30			234	1.11	-0.02	0.709090	0.000028

therefore, no significant vertical flow down through the sediments appears to be taking place. The actual geothermal gradient at Site 817 was measured in the sediments using the WSTP tool and found to be 57°C/km. In contrast, at Site 825/811, the gradient was slightly higher at 80°C/km.

In conclusion, although the temperatures obtained during logging do suggest the presence of over- and underpressured zones in the sediments, these data do not reflect the geothermal gradients actually present in the sediments and, therefore, cannot be taken to be indicative of water flow. The slight differences between Sites 825 and 817 indicate the presence of different geothermal gradients, which may be sufficient to induce circulation. To further confirm the presence of geothermally driven water flow, an extensive heat flow survey should be conducted over the Queensland Plateau.

Deep Trough Sites: 815 and 823

The overall geochemistry of Sites 815 and 823 is dominated by: (1) intense sulfate reduction and carbonate recrystallization in the upper portion of the sediment column; and (2) the increasing salinity of the pore fluids with increasing depth. At Site 815, the concentration of Ca^{2+} remains constant throughout the upper 200 mbsf. Throughout this same interval, the concentration of Mg²⁺ decreases to 26 mM, Sr²⁺ increases to 1400 μ M, and SO₄²⁻ decreases to 2 mM. Below 200 mbsf, the concentration of all components, with the exception of Sr²⁺, increases to the bottom of the core (473 mbsf). The decrease in Sr²⁺ occurs because of the increase in SO₄²⁻, which in turn causes the mineral celestite to become supersaturated. In contrast, Site 823 penetrated 1103 mbsf and exhibits much more pronounced decreases in Ca²⁺, Mg²⁺, and K⁺. Site 823 also contains a much lower carbonate content and, consequently, a higher concentration of clay minerals. Site 823 also exhibits an increase in the salinity of the pore fluids in the lower portions of the hole, but at a reduced rate compared with that at Site 815. The change in δ^{18} O from –0.5 to +0.5 is much greater at Site 823 than at Site 815, which possesses heavy oxygen isotopic values throughout.

It has been suggested that the general increase in chlorinity at Sites 815 and 823 resulted from the diffusion of salts from an underlying evaporite unit (Davies, McKenzie, Palmer-Julson, et al., 1991). Increases in Cl⁻ also were observed at Sites 817, 818, and 822 (Fig. 12). Such a common trend suggests that perhaps the evaporitic unit postulated to underlie Sites 815 and 823 may be an extensive feature of the Queensland Trough. Such a deposit has not as yet been identified in the Townsville Trough, but is known in sediments of Oligocene age in the Capricorn Basin approximately 100 km to the south (Ericson, 1976).



Figure 3. Oxygen isotopic data from pore waters recovered from Sites 819, 820, 821, 822, and 823.

Diffusion from an underlying saline source is neither supported nor refuted by stable O and C isotopic data. At Site 815, the δ^{18} O is isotopically enriched throughout the entire core and shows no significant changes in the lower portion of the core corresponding to the salinity increase. The δ^{13} C exhibits a slight increase in the lower samples. At Site 823, the δ^{18} O shows an increase in the lower portion, although this cannot be definitively ascribed to the chlorinity changes and may be a result of recrystallization at higher temperatures.

Site 823 exhibits perhaps the highest concentration of Sr^{2+} yet measured in pore waters from either the ODP or DSDP drilling. Concentrations as high as 2774 μ M are reached at 934.6 mbsf. These fluids are still below celestite saturation, as the sulfate has been completely depleted. These high values are reached as a result of the relatively low permeability of the sediments, combined with the

absence of sulfate and the continued recrystallization of the carbonate sediments.

Great Barrier Reef Margin Sites

Sites 822, 819, 820, and 821 form a distal-to-proximal transect off the eastern margin of the Queensland continental margin. The sediments are composed primarily of mixtures of periplatform carbonates derived from the Great Barrier Reef and quartz, feldspars, and clay minerals eroded from the Australian mainland. The concentration of clay minerals increases with increasing distance from the continental shelf, making up more than 50% of the sediments at Sites 819 and 820. Quartz usually composes between 10% and 20% of the sediments at each site, although the proximal sites have a slight tendency



Figure 4. Oxygen and carbon isotopic data from pore waters from Site 823.

to contain higher concentrations. The clay minerals mainly consist of kaolinite and illite, and the feldspars are principally albite.

Calcium, Magnesium, Potassium, Strontium, and Oxygen Isotopes

Decreases in the concentrations of K⁺ and Mg²⁺ and increases in Na+ concentration were attributed to the diagenesis of igneous minerals (Davies, McKenzie, Palmer-Julson, et al., 1991). In addition, data presented in this study show a depletion in δ^{18} O in the pore fluids. In previous work, depletions in the concentrations of K⁺, Mg²⁺, and δ^{18} O, as well as increases in Ca²⁺, have been attributed to alteration of Layer 2 basalts or volcanic material within the sediments to clay minerals or zeolites at elevated temperatures (Lawrence et al., 1976; Perry et al., 1976). As the Leg 133 sites are not located immediately over basement, reactions responsible for the depletion in Mg2+, K+, and δ^{18} O and the decrease in Ca²⁺ must be taking place within the sediments (Fig. 13). Because the maximum temperature of the sediments cored is probably less than 50°C, it is unlikely that the alteration of clay minerals, as originally suggested (Davies, McKenzie, Palmer-Julson, et al., 1991), can be invoked to account for the geochemical changes observed in the pore waters. One possible explanation for the decrease in K⁺ and increase in Na⁺ involves absorption and desorption from illite (Hoffman, 1979). Although changes in the δ^{18} O of the pore waters then could be explained by the alteration of volcanic glass, the observed decrease also could be a result of recrystallization of the relatively isotopically light precursor carbonate in colder waters. For example, Figure 14 shows the evolution of pore water in a closed system as it is subject to carbonate recrystallization. The equations used to construct these plots have been taken from Killingley (1983). In each of the three cases, the isotopic composition of the initial water has been set at +0.5‰ standard mean ocean water (SMOW), the bottom temperature at 10°C, and the geothermal gradient at 57°C/km (data taken from Davies, McKenzie, Palmer-Julson, et al., 1991). It



Figure 5. Oxygen and carbon isotopic data recovered from pore waters at Site 815.

is assumed that 100% recrystallization occurs by 400 mbsf and that once carbonate has been recrystallized it no longer reacts with the pore fluid. The difference among the three lines in Figure 14 represents differing isotopic compositions of the precursor. As may be observed, each case shows a depletion with depth. These calculations are in some ways similar to the measured data shown in Figure 3. Obviously, some discrepancies exist, such as the elevated isotopic composition at Site 821. This may be a result of fossil water trapped in the sediment. In addition, Site 821 exhibits the steepest Sr2+ gradients and, hence, the highest amount of carbonate recrystallization. However, this site does not exhibit as significant a depletion in the δ^{18} O as at Sites 819 and 822, which on the basis of their Sr²⁺ gradients should show lower rates of recrystallization. Explanation for these differences may lie in the different isotopic composition of the precursor, differences in the isotopic composition of the water, differences in the rate of recrystallization, and even some degree of alteration of volcanic glasses.

All the continental margin sites have heavy δ^{18} O values in the upper 50 to 100 mbsf. We think that these heavy values are related to the original isotopic composition of the pore fluids formed on the adjacent continental shelf area.

Oxidation of Organic Material

The rapid rate of sedimentation at Sites 819 to 822 (100 to 200 m/m.y.) has resulted in the burial of organic material that has been subsequently oxidized by sulfate-reducing bacteria to produce H_2S and CO_2 . In turn, these gases have been instrumental in the precipitation of pyrite, dolomite, and calcium carbonate. As the types and



Figure 6. Carbon isotopic data from pore waters at Sites 811 through 814.

amount of minerals and organic material vary with distance from the shelf, the nature and magnitude of the diagenetic reactions occurring in the sediments also change, producing distinctive differences in the profiles of the interstitial minor elements at the four sites.

The most dramatic change visible at all sites is the removal of sulfate during the processes of organic material oxidation (Fig. 7). Although complete sulfate reduction occurs at all sites, it takes place as shallow as 25 mbsf at Sites 822 and 819, while at Site 821, sulfate is not completely exhausted until 144.5 mbsf. Small amounts of sulfate detected in the pore fluids below the sulfate depletion zone are thought to be a result of contamination. As all sites have experienced approximately similar rates of sedimentation and do not exhibit significant differences in their physical properties (permeability, density etc.), the most plausible explanation for the different depths of sulfate reduction lies in the different amounts and types of organic material present at each site. Sites 819 and 822 must have initially contained higher concentrations of organic matter or material that was more easily degraded than that at Sites 821 and 820. This initial difference is reflected in the present content of organic material at these sites; the organic material increases with increasing distance from the continental shelf. A further possibility is that, because the organic material at Sites 819 and 822 is predominantly of a marine rather than a terrestrial origin, it is more easily degraded by sulfatereducing bacteria. Such a difference would promote higher rates of sulfate reduction at the distal sites. The difference in the rate of sulfate reduction is partially reflected in the variations in alkalinity, with the highest values occurring at Sites 819 and 822. The increase in alkalinity at these sites, however, is significantly less than that which might be expected as a result of sulfate reduction alone. This alkalinity deficit can be explained by: (1) the precipitation of dolomite and calcite; (2) loss of H_2S during the precipitation of pyrite.

Beneath the zone of sulfate reduction, large increases occur in the concentrations of methane. It has been suggested that these changes in the concentration might influence the $\delta^{13}C$ of the DIC and the consequent isotopic composition of diagenetic calcite and dolomite. However, although there were small depletions in the $\delta^{13}C$ of the DIC, there were no large changes as might have been expected based on the theories of Irwin et al. (1977) (Fig. 7). In fact, the most negative values in the sulfate reduction zone occurred at Site 819, and the smallest depletions at Site 821. It has been suggested that the absence of negative isotopic compositions resulted from the system being heavily buffered by carbonate minerals (Swart, this volume).

The magnitude of the increase in the concentration of Sr^{2+} in the interstitial fluids is often considered to be an indicator of the degree of carbonate recrystallization. If this is the case, then the amount of carbonate crystallization increases from the distal toward the proximal sites (Fig. 9). This observation supports the interpretation of the carbon isotopic data, which shows the smallest depletion in δ^{13} C at the proximal sites in the sulfate reduction zone and the smallest enrichment in the methanogenesis zone. Hence, the greater depletion in δ^{13} C seen at the distal sites and the higher isotopic compositions attained in the lower portions suggest that carbonate recrystallization is less important at these sites.

Accompanying these changes in pore-water chemistry is an increase in the concentration of dolomite from the distal to proximal sites (Davies, McKenzie, Palmer-Julson, et al., 1991). Therefore, the sites having the greatest degree of sulfate reduction and the highest alkalinity have the lowest concentrations of dolomite. Such a distribution is contrary to current hypotheses (Baker and Kastner, 1981) that dolomitization may be favored by high concentrations of alkalinity and low amounts of sulfate (Swart, this volume).

CONCLUSIONS

Pore fluids from Leg 133 can be separated into three groups on the basis of the processes that are postulated to control their composition.

1. Pore fluids from the Queensland Plateau sites are characterized by an absence of geochemical gradients, by modern strontium isotopic ratios, and, in some instances, by unusual geothermal gradients. These suggest the presence of significant fluid flow through the Queensland Plateau. The absence of a depletion in the δ^{18} O value of the pore waters and the slightly negative δ^{13} C values support the interpretation of fluid movement. The deeper flank sites (Sites 817 and 818) do not show evidence of fluid movement through the upper periplatform portion of the section. However, at Site 817, an underpressured zone was drilled that caused water to be drawn into the hole during drilling, suggesting the penetration of a fluid conduit.

2. The deeper trough sites (Sites 815 and 823) have exhibited diffusive control of pore-water constituents, combined with significant amounts of carbonate recrystallization. In the lower portion of the section, chloride concentrations increased together with the concentrations of various other constituents. These increases suggest the presence of an underlying evaporite unit. Such increases also were visible in the lower portions of Sites 817 and 818.

3. The Queensland continental margin sites are characterized by increasing rates of carbonate mineral diagenesis toward the Great Barrier Reef. In spite of total sulfate reduction and high rates of methanogenesis at these sites, little evidence could be ascertained of these processes in the δ^{13} C of the pore waters or in the isotopic composition of the diagenetic minerals. The isotopic signatures of sulfate reduction and methanogenesis were more evident in the distal sites. Although significant depletion in the Ca²⁺, Mg²⁺, K⁺, and δ^{18} O content of the pore fluids was seen, only a decrease in K⁺ and an

increase in Na⁺ can be ascribed to interaction with clay minerals or formation of clay minerals from volcanic glasses.

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REFERENCES*

- Baker, P.A., and Kastner, M., 1981. Constraints on the formation of sedimentary dolomite. *Science*, 213:215–216.
- Davies, P.J., McKenzie, J.A., Palmer-Julson, A., et al., 1991. Proc. ODP, Init. Repts., 133: College Station, TX (Ocean Drilling Program).
- Ericson, E.K., 1976. Capricorn Basin. In Leslie, R.B., Evans, H.J., and Knight, C.L. (Eds.), Economic Geology of Australia and Papua New Guinea (Vol. 3). Australas. Inst. Mining and Metall. Monogr., 7:446–450.
- Gieskes, J.M., Elderfield, H., and Palmer, M.R., 1986. Strontium and its isotopic composition in interstitial waters of marine carbonate sediments. *Earth Planet. Sci. Lett.*, 77:229–235.
- Hoffman, J.C., 1979. An evaluation of potassium uptake by Mississippi-Riverborne clays following deposition in the Gulf of Mexico [Ph.D. dissert.]. Case Western Reserve Univ., Cleveland, OH.
- Irwin, H., Curtis, C.D., and Coleman, M.L., 1977. Isotopic evidence for the source of diagenetic carbonates formed during burial of organic-rich sediments. *Nature*, 269:209–213.

- Killingley, J.S., 1983. Effects of diagenetic recrystallization of ¹⁸O/¹⁶O values of deep-sea sediments. *Nature*, 301:594–597.
- Lawrence, J.R., Gieskes, J.M., and Anderson, T.F., 1976. Oxygen isotope material balance calculations, Leg 35. In Hollister, C.D., Craddock, C., et al., *Init. Repts. DSDP*, 35: Washington (U.S. Govt. Printing Office), 507–512.
- Perry, E.A., Jr., Beckles, E.C., and Newton, R.M., 1976. Chemical and mineralogical studies, Sites 322 and 325. *In* Hollister, C.D., Craddock, C., et al., *Init. Repts. DSDP*, 35: Washington (U.S. Govt. Printing Office), 465–469.
- Schlager, W., and James, N.P., 1978. Low-magnesian calcite limestones forming at the deep-sea floor, Tongue of the Ocean, Bahamas. *Sedimentology*, 25:675–702.
- Swart, P.K., and Burns, S.J., 1990. Pore-water chemistry and carbonate diagenesis in sediments from Leg 115: Indian Ocean. *In Duncan*, R.A., Backman, J., Peterson, L.C., et al., *Proc. ODP, Sci. Results*, 115: College Station, TX (Ocean Drilling Program), 629–645.
- Swart, P.K., and Guzikowski, M., 1988. Interstitial-water chemistry and diagenesis of periplatform sediments from the Bahamas, ODP Leg 101. In Austin, J.A., Jr., Schlager, W., Palmer, A.A., et al., Proc. ODP, Sci. Results, 101: College Station, TX (Ocean Drilling Program), 363–380.

Abbreviations for names of organizations and publications in ODP reference lists follow the style given in *Chemical Abstracts Service Source Index* (published by American Chemical Society).

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Figure 7. Sulfate, alkalinity, and carbon isotopic data from pore waters at Sites 819 through 822.



Figure 8. A comparison of the changes in the concentration of Sr^{2+} at Site 631 (Leg 101, Swart and Guzikowski, 1986), Site 716 (Leg 115, Swart and Burns, 1988), and Site 812 (Leg 133). Note the absence of an increase in Sr^{2+} at Site 812 compared with the other two sites.



Figure 9. Interstitial pore-water calcium and strontium gradients from Sites 812 through 814.

⁸⁷ Sr / ⁸⁶ Sr



Figure 10. Strontium isotopic data from Sites 812 through 814 compared to contemporaneous seawater curve (gray shading). Note that most values are more radiogenic than contemporaneous seawater and are close to modern seawater values, suggesting circulation through the sediments (data from Elderfield et al., this volume). Dissolved Sr^{2+} profiles are shown in Figure 9.



Figure 11. A. Strontium isotopic data from Sites 817 and 823 compared to contemporaneous seawater curve (data from Elderfield et al., this volume). Also shown are data from Figure 9 for comparison. Strontium isotopic data fall below the seawater curve, in contrast to data from Sites 812 to 814. **B.** Strontium concentration data from Sites 817 and 823.



Figure 11 (continued).



Figure 12. The concentrations of Cl^{-} at Sites 815, 817, 818, and 823. All sites exhibit a general increase in the lower portions of the hole.



Figure 13. Changes in the concentrations of K^+ , Mg^{2+} , Sr^{2+} , and Ca^{2+} at Sites 819 through 822.



