8. CR-RICH SPINEL AS A TRACER FOR MELT MIGRATION AND MELT–WALL ROCK INTERACTION IN THE MANTLE: HESS DEEP, LEG 147¹

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ABSTRACT

Site 895 of the Ocean Drilling Program Leg 147 cored melt-impregnated peridotite from the southeast flank of the Hess Deep intrarift high, recovering generalized stratigraphic sequences of harzburgite-dunite-troctolite-gabbro-troctolite-dunite-harzburgite, with the gabbros representing fossil melt conduits through preexisting harzburgite. Cr-rich spinel within the Site 895 samples reflects and records the effects of these multiple injections of melt into depleted, residual harzburgites of the East Pacific Rise uppermost mantle. Gabbroic spinel compositions vary widely (TiO₂, 0.55%–2.56%; Mg/[Mg + Fe²⁺], 0.30–0.60), showing that the impregnating melts became locally quite evolved, with variation in melt composition controlled by both crystal fractionation and local melt-rock interaction. TiO₂ contents in associated dunitic spinel are 10–20 times as high as in harzburgitic spinels (0.5%–1.1% vs. $\leq 0.07\%$, respectively), suggesting that the dunite was produced by melt–wall rock interaction associated with melt impregnation. This process produced halos of dunite around residual bodies of gabbroic rock and possible residual dunite "trails" of melt migration through peridotite. Spinel and other phase compositional data are consistent with the injection of depleted, exotic N-MORB melt by both diking and intergranular flow at depths shallower than 10 km and temperatures well above 900°C, and most likely approaching about 1230°C. High ZnO (to 0.75%) in the metamorphically altered rims of spinel in gabbro in cotolite indicates that Zn-rich hydrothermal fluids may have been present during post-crystallization, lower amphibolite to zeolite facies metamorphic alteration of the Hess Deep peridotites, with the gabbroic melt impregnations providing possible preferential conduits for fluid flow.

INTRODUCTION

Processes of melt and fluid generation, transportation, storage, and crystallization in the oceanic upper mantle and crust are at present imperfectly understood, with progress dependent on recovery of samples that preserve stratigraphic relations. Leg 147 of the Ocean Drilling Program (ODP) represented the first leg planned around the strategy of "offset drilling," designed to create a composite section of oceanic crust and uppermost mantle by drilling a number of holes in crustal sections exposed by faulting. During Leg 147, Sites 894 and 895 were cored on the summit and southeast flank of the Hess Deep intrarift high, a north-tilted fault block within Hess Deep, a 2500-mdeep fault basin at the tip of the westward-propagating Cocos-Nazca spreading center (Fig. 1; Lonsdale, 1988; Hekinian et al., 1993; Francheteau et al., 1990; Gillis, Mével, Allan, et al., 1993). Drilling at these sites recovered sections of mid-crustal gabbroic rocks and uppermost mantle peridotite, respectively (Gillis, Mével, Allan, et al., 1993).

The recovery from the six Site 895 holes consists of peridotite that has been impregnated by basaltic melt at a depth most likely near the crust/mantle boundary (Gillis, Mével, Allan, et al., 1993; Fig. 2). Although rocks of similar nature have been recovered from the Hess Deep (Hekinian et al., 1993; Girardeau and Francheteau, 1993), the Site 895 samples are of greater petrological value owing to their fresher nature and to the preservation of stratigraphic relations within the core. Although complicated by imperfect recovery and hole location in what might be rotated fault slumps (Kelso et al., this volume; Kikawa et al., this volume), the recovered core nevertheless shows fairly consistent local stratigraphic assemblages of harzburgite-dunite-troctolite-gabbroic rock-troctolite-dunite-harzburgite. The intimate association of large amounts of gabbroic rock with dunite in core is similar to what has been observed in ophiolites (Nicolas, 1989) and is exemplified by the comparative recoveries of Holes 895D and 895E (Fig. 3). The gabbroic and troctolitic rocks represent melt injection, by diking and intergranular flow (perhaps on dike margins), into host peridotite at shallow mantle levels (Gillis, Mével, Allan, et al., 1993; Hekinian et al., 1993; Girardeau and Francheteau, 1993; Dick, 1989; Nicolas, 1989). The sporadic presence of contact relations within the gabbroic rocks (Fig. 4, showing an olivine gabbro crosscutting a troctolite) underscores the interpretation of the Leg 147 shipboard party that melt injection into these peridotites occurred in multiple events (Gillis, Mével, Allan, et al., 1993).

Cr-rich spinel is a ubiquitous phase in nearly all Site 895 rock. It is a highly sensitive indicator of a wide variety of petrogenetic processes (Dick and Bullen, 1984; Allan et al., 1988), recording variations in host melt and coexisting phase chemistry, temperature, and fO₂. In this paper, we use high-precision analyses of spinel within these samples to track processes of mass transfer during basalt melt impregnation, and use this data to address two of the most vexing problems of mid-ocean-ridge basalt (MORB) petrogenesis: the nature of melt transport and the degree of melt–wall rock interaction in the upper mantle beneath spreading ridges. We focus on reporting spinel core-to-rim transects of major and minor element variation, examining the effects of basaltic melt–wall rock interaction on both the impregnating melt and the surrounding peridotite wall rock, exploring the origin of mantle dunite, and placing constraints on the physical conditions of melt impregnation.

¹Mével, C., Gillis, K.M., Allan, J.F., and Meyer, P.S. (Eds.), 1996. Proc. ODP, Sci. Results, 147: College Station, TX (Ocean Drilling Program).

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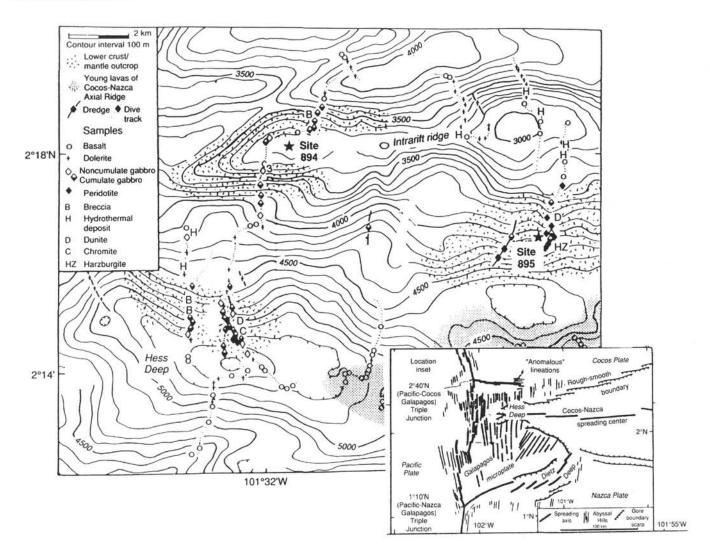


Figure 1. Location map of Hess Deep sites cored during Leg 147 (from Gillis, Mével, Allan, et al., 1993). The geology shown is modified from Francheteau et al. (1990).

PETROGRAPHY

For all sample descriptions, ODP conventions (Gillis, Mével, Allan, et al., 1993) are followed concerning sample identification (leg, core, core type, section, interval) and alteration terminology (fresh, 0%-2%; slight, 2%-10%; moderate, 10%-40%; high, 40%-80%; pervasive, >80\%). Extensive petrographic descriptions of these samples are given in Gillis, Mével, Allan, et al. (1993), some of which are the actual thin sections analyzed here, so only a brief summry of the sample petrography will be given.

Harzburgites recovered at Site 895 are typical of more depleted, clinopyroxene-poor, residual harzburgites (Dick, 1989; Dick et al., 1984; Dick and Fisher, 1984), with an original protogranular texture that has been overprinted with porphyroclastic textures and other subsequent textures associated with brittle deformation and hydro-thermal alteration. They contain 10%–20% porphyroclasts of orthopyroxene (to about 5 mm), partially altered to talc, tremolite, and bastite (Früh-Green, Plas, and Lecuyer, this volume; Mével and Stamoudi, this volume), and only minor amounts (typically <1%) of small (<1 mm), anhedral clinopyroxene, commonly attached to spinel. Individual olivine grains are up to 1 cm in size, although they are commonly represented by relict "eyes" surrounded by serpentine alteration. Dark, reddish-brown spinel (to about 5 mm) occurs in a

variety of morphologies, ranging from blocky anhedral (Pl. 1A) to anhedral "holly-leaf" shapes (Pl. 1B), with vermicular forms less common but present. Although much fresher than typical abyssal peridotites and fresher than the harzburgites from Hole 895E, the analyzed harzburgites from Holes 895A, 895C, and 895D still carry about 50%–80% alteration minerals.

The original protogranular texture of the harzburgites is interpreted to be associated with asthenospheric flow; the porphyroclastic textures are interpreted to be associated with later, subsolidus flow, whereas the hydrothermal and brittle deformation textures are thought to be related to the Hess Deep opening (Boudier et al., this volume; Dick and Natland, this volume; Kennedy et al., this volume). The major and trace element mineral compositions of the harzburgites lie at the extreme depleted end of the field for residual harzburgites worldwide (Dick and Natland, this volume), but do not extend to the extremely depleted compositions found in many ophiolitic peridotites (Dick and Fisher, 1984).

Dunite samples from Site 895 contain little or no fresh pyroxene, and are notable in that they are substantially more altered than the associated harzburgites, exhibiting extensive to pervasive alteration of their olivine to serpentine. Bastite pseudomorphs of orthopyroxene are present in small amounts (0%–10%; Gillis, Mével, Allan, et al., 1993). Spinel within the analyzed dunites is dark reddish-brown to

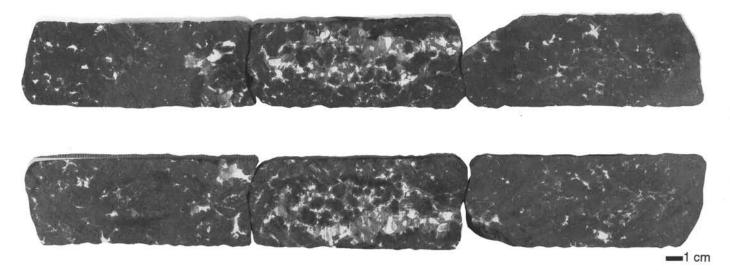


Figure 2. Photograph of core showing pervasive melt impregnation into peridotite. Harzburgite occurring at the top of Core 147-895C-4R grades into orthopyroxene-poor dunite, which in turn is intimately invaded by intergranular gabbroic rock. Near and within the dunite, visible at the top of the photograph, are interstitial, irregular patches of anhedral plagioclase, clinopyroxene, olivine, and Cr-spinel that increase in size and abundance down the core to the center of the photograph. The melt patches are overprinted with lower amphibolite to zeolite facies metamorphic minerals. Troctolite samples analyzed in this paper come from the margins of the interval shown here.

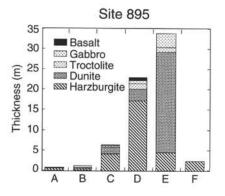


Figure 3. Comparative rock recoveries from the six Site 895 holes, shown in the recorded thickness of the recovered rocks. Note the relative abundance of both gabbroic rocks and dunite at Hole 895E.

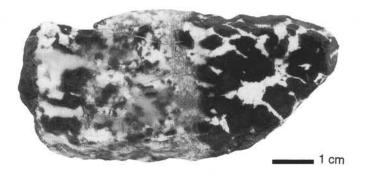


Figure 4. A rare contact recovered in core between troctolite and an apparently later gabbroic intrusion. Note the chilled margin on the gabbro.

black, and is typically more rounded in shape than those in the harzburgites, with anhedral, holly-leaf textures less well developed (Pl. 1C, D). Spinel grains commonly occur in elongate trains, with these trains also occurring in harzburgite and most likely representing a foliation fabric (Gillis, Mével, Allan, et al., 1993).

Troctolites from Hole 895C are named on a modal basis and represent an intimate patchwork of host dunite pods and impregnating gabbroic veins and patches (Fig. 3). Typically, the ratio of gabbroic rock to dunitic pods is less than 1; these samples could alternatively be called plagioclase dunites, depending on the amount of plagioclase present. The gabbroic segregations consist predominantly of anhedral plagioclase and clinopyroxene that have been heavily altered to secondary amphibolite to zeolite minerals (Gillis, Mével, Allan, et al., 1993; Früh-Green, Plas, and Lecuyer, this volume; Mével and Stamoudi, this volume). They are typically irregular in shape, with their edges defined by the heavily altered, surrounding olivine kernels (Pl. 1E). Dark, nearly opaque spinel is abundant (to 5% of the samples analyzed here), and occurs in two forms: blocky, subhedral grains to several millimeters in size (Pl. 1F), and large (to 1 cm), skeletal grains that typically include plagioclase and unexsolved clinopyroxene (Pl. 1F). These spinel grains may occur as crystals within or on the margins of gabbroic patches (Pl. 1E), or they may occur in isolation, surrounded by olivine (Pl. 1F).

Olivine gabbro in the samples analyzed contains a much lower percentage of olivine. Where fresh (such as in Sample 147-895D-7R-2, 71–75 cm; Pl. 1G), olivine occurs in isolated clusters of millimetersized grains, surrounded by granular and intergranular clinopyroxene and plagioclase that is partially to completely altered to amphibolite to zeolite metamorphic minerals (Gillis, Mével, Allan, et al., 1993; Früh-Green, Plas, and Lecuyer, this volume; Mével and Stamoudi, this volume). Spinel is dark and nearly opaque, occurring as small (0.05–0.4 mm), commonly rounded, subhedral to euhedral grains or inclusions in olivine and plagioclase (Pl. 1G, H). Other gabbroic samples lack fresh olivine and usually occur farther away from dunite in recovered core; these are typically very highly to pervasively altered to secondary minerals, with spinel occurring as small (<0.3 mm), surprisingly fresh, isolated euhedra. Spinel abundance in these samples (Samples 147-895D-10W-1, 82–85 cm, 147-895D-8R-2, 42–44 cm, and 147-895E-1R-1, 109–112 cm, analyzed here) is much lower than in the other, more olivine-rich gabbro samples.

ANALYTICAL METHODS

A major goal of this study was to measure accurately a variety of minor elements (including Ti, V, Zn, Co, and Sc) in the spinels to use as indicators of mass transfer processes. An additional goal was to make high-quality estimations of spinel Fe₂O₃ and Fe³⁺/(Cr + Al + Fe³⁺) (Carmichael, 1967), estimations that require complete elemental analysis in order to be successful. Electron microprobe analyses of spinel were obtained on a Cameca SX-50 microprobe using Smithsonian natural mineral standards (Jarosowich et al., 1980) and pure metal standards, a focused beam, and a sample current of 30 nA. Great care was taken to avoid overt alteration in the spinel grains, with back-scattered electron images taken of each grain before the analysis location was planned (Fig. 5). Unusually long counting times (60-180 s/element) were utilized for all elements except for Fe. Ca, and Si, which were counted for 30, 30, and 10 s, respectively. Special care was also taken in placing where peak and background counts were taken in order to avoid peak overlap problems. Nevertheless, peak overlap corrections, using interference counts made on highly purified metal and rutile, were made for correction of interference of Cr on Mn, Ti on V, and Fe on Co. Unfortunately, the analytic procedure employed was not able to measure Co and Sc reliably, for which these oxides are estimated at ≤0.02% abundance in the Hess Deep spinels. Results of these analyses are presented in Tables 1 and 2 and Figures 6 through 8. It is important to note that Cr# in this paper relates to $Cr/(Cr + Al + Fe^{3+})$ rather than Cr/(Cr + Al) in order to note the composition of Cr relative to all of the major trivalent spinel cations. Selected grains of coexisting olivine, pyroxene, and plagioclase were also analyzed on the electron microprobe, using a focused beam, sample currents of 10-30 nA, counting times of 10 to 60 s, and both natural mineral (Jarosowich et al., 1980) and metal standards. These analyses are presented in Tables 3 and 4.

SPINEL CHEMISTRY

Spinel Metamorphic Overprint

The Leg 147 Hess Deep peridotites are composite rocks, originating from a combination of igneous processes and high-temperature metamorphic processes in the upper mantle. They are progressively overprinted with the effects of (1) high-temperature (>500°C), nearridge amphibolite facies metamorphism; (2) dominant, lower temperature (350-450°C), off-ridge greenschist facies metamorphism that resulted in extensive serpentinization; and (3) subsequent low-temperature zeolite alteration (Früh-Green, Plas, and Lecuyer, this volume; Mével and Stamoudi, this volume; Gillis, Mével, Allan, et al., 1993). In addition, some amounts of Mg-Fe²⁺ exchange between spinel and olivine during peridotite cooling before amphibolite-zeolite metamorphism has occurred (e.g., Ozawa, 1983). As a result, interpretation of spinel phase chemistry, including the understanding of basic elemental variation and covariation, is difficult. Our principal interest is to examine variations in spinel composition that result from processes associated with melt impregnation and melt-wall rock interaction, which is a difficult task without removal of the effects of the lower temperature metamorphic overprint.

Each analyzed spinel was imaged by back-scattered electrons in order to avoid analyzing regions of obvious ferritchromite formation that commonly occur along grain boundaries and crystal fractures (Fig. 5). Despite this precaution, examination of the electron microprobe analytical transects shows that many of the analyses of spinel reflect compositional effects of metamorphic exchange after initial crystallization, as exemplified by slight to sharp lowering in Mg/(Mg

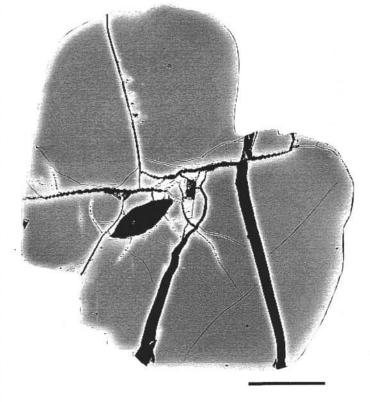


Figure 5. Back-scattered electron image of a spinel from Sample 147-895E-2R-1, 36–39 cm. This is the same grain shown by analytical transect A1 in Figure 6; it is also shown in Plate 1H.

+ Fe²⁺) (Mg#), which is typically accompanied by a loss in Ti at the grain margins (Fig. 6). Severe lowering in Mg# at spinel margins, caused by Mg loss and Fe²⁺ gain (such as in the electron microprobe transects of Sample 147-895E-2R-1, 36–39 cm, shown in Fig. 6 and by the associated analyses A1-IPC, A1-IPR, B1-IPC, and B1-IPR in Table 2), is usually accompanied by increases in ZnO and MnO. Similar increases in MnO occur in spinel from alpine peridotites affected by low-temperature alteration (S. Arai, pers. comm., 1994); the increase in ZnO (up to 0.75%) probably reflects exchange with Zn-rich hydrothermal fluids.

A key compositional feature of this metamorphic overprint is the lack of any significant change in Cr#, even with a greater than fourfold decrease in Mg#. This cannot possibly represent an igneous process related to exchange with an evolving melt, owing to the strong compositional coupling between Fe²⁺ and Cr³⁺ and Mg²⁺ and Al³⁺ in the spinel structure (Sack, 1982; Sack and Ghiorso, 1991a, 1991b). This is illustrated by examining the free energy of exchange (ΔG_{EX}) for the appropriate reactions:

$$MgCr_2O_4 + FeAl_2O_4 \leftrightarrow MgAl_2O_4 + FeCr_2O_4$$
 (1)

 $G_{EX} \approx -23 \text{ kJ Mol}^{-1}$ (Sack and Ghiorso, 1991a, 1991b)

in the solid spinel phase, and

$$(Mg^{2+})_{Melt} + (Fe^{2+})_{Spinel} \leftrightarrow (Mg^{2+})_{Spinel} + (Fe^{2+})_{Melt}$$
(2)

 $\Delta G_{EX} \approx -13 \text{ kJ Mol}^{-1}$ (Allan et al., 1988; Allan, 1992).

These relations demand that during equilibrium exchange between spinel and melt, large drops in Mg# must be accompanied by

Table 1. Summary of Cr-spinel electron microprobe analyses.

bbro 66 bbro 76 bbro 20 28 30 30 85 62 e 9 e 30	0-93.7 66.8 76.1 None 18 32 1.1 20.0 2 1.1 20.0 10 28.5 30.2 6 30.2 30.2 6 30.7 6 85.0 42 62.4 43 9.3 None 30.0 5	0.81-1.04 0.92-2.56 0.52-1.21 0.71-1.16 0.92-1.77 1.39-1.84 0.84-1.52 0.72-1.14 0.44-0.48 0.90-1.10 0.05-0.07	0.27-0.32 0.16-0.65 0.20-0.34 0.18-0.38 0.32-1.01 0.30-0.42 0.26-0.32 0.23-0.30 0.17-0.23 0.21-0.28 0.18-0.25	0.24-0.39 0.26-0.62 0.17-0.47 0.23-0.30 0.24-1.11 0.29-0.63 0.23-0.38 0.26-0.35 0.23-0.33 0.14-0.26	0.07-0.15 0.04-0.15 0.09-0.15 0.05-0.14 0.13-0.21 0.13-0.20 0.13-0.20 0.11-0.17 0.12-0.19	0.13-0.20 0.07-0.38 0.06-0.21 0.08-0.15 0.04-0.75 0.17-0.43 0.08-0.22 0.08-0.19 0.08-0.24	0.05-0.09 0.09-0.15 0.07-0.08 0.04-0.07 0.05-0.12 0.10-0.14 0.07-0.11 0.07-0.09 0.05-0.07	$\begin{array}{c} 0.51-0.58\\ 0.35-0.54\\ 0.50-0.53\\ 0.44-0.52\\ 0.45-0.60\\ 0.48-0.55\\ 0.49-0.55\\ 0.46-0.53\\ 0.47-0.50\\ \end{array}$	$\begin{array}{c} 0.42 - 0.51 \\ 0.25 - 0.50 \\ 0.32 - 0.61 \\ 0.43 - 0.49 \\ 0.13 - 0.58 \\ 0.29 - 0.49 \\ 0.40 - 0.56 \\ 0.51 - 0.60 \\ 0.50 - 0.61 \end{array}$	10 45 24 8 12 10 41 20 20			
bbro 66 bbro 76 bbro 20 28 30 30 85 62 e 9 e 30	66.8 18 76.1 32 1.1 2 20.0 10 28.5 6 30.2 6 30.7 6 85.0 42 62.4 43 9.3 None	0.92-2.56 0.52-1.21 0.71-1.16 0.92-1.77 1.39-1.84 0.84-1.52 0.72-1.14 0.44-0.48 0.90-1.10	0.16-0.65 0.20-0.34 0.18-0.38 0.32-1.01 0.30-0.42 0.26-0.32 0.23-0.30 0.17-0.23 0.21-0.28	0.26-0.62 0.17-0.47 0.23-0.30 0.24-1.11 0.29-0.63 0.23-0.38 0.26-0.35 0.23-0.33	0.04-0.15 0.09-0.15 0.05-0.14 0.13-0.21 0.13-0.20 0.13-0.20 0.11-0.17	0.07-0.38 0.06-0.21 0.08-0.15 0.04-0.75 0.17-0.43 0.08-0.22 0.08-0.19	0.09-0.15 0.07-0.08 0.04-0.07 0.05-0.12 0.10-0.14 0.07-0.11 0.07-0.09	0.35-0.54 0.50-0.53 0.44-0.52 0.45-0.60 0.48-0.55 0.49-0.55 0.46-0.53	0.25-0.50 0.32-0.61 0.43-0.49 0.13-0.58 0.29-0.49 0.40-0.56 0.51-0.60	45 24 8 12 10 41 20			
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30 30 85 62 e 9 e 30	30.2 6 30.7 6 85.0 42 62.4 43 9.3 None	0.84-1.52 0.72-1.14 0.44-0.48 0.90-1.10	0.26-0.32 0.23-0.30 0.17-0.23 0.21-0.28	0.23-0.38 0.26-0.35 0.23-0.33	0.13-0.20 0.13-0.20 0.11-0.17	0.08-0.22 0.08-0.19	0.07–0.11 0.07–0.09	0.49–0.55 0.46–0.53	0.40-0.56 0.51-0.60	41 20			
30 30 85 62 e 9 e 30	30.2 6 30.7 6 85.0 42 62.4 43 9.3 None	0.84-1.52 0.72-1.14 0.44-0.48 0.90-1.10	0.26-0.32 0.23-0.30 0.17-0.23 0.21-0.28	0.23-0.38 0.26-0.35 0.23-0.33	0.13-0.20 0.13-0.20 0.11-0.17	0.08-0.22 0.08-0.19	0.07–0.11 0.07–0.09	0.49–0.55 0.46–0.53	0.40-0.56 0.51-0.60	41 20			
3(85 62 e 9 e 3(30.7 6 85.0 42 62.4 43 9.3 None	0.72-1.14 0.44-0.48 0.90-1.10	0.23-0.30 0.17-0.23 0.21-0.28	0.26-0.35 0.23-0.33	0.13-0.20 0.11-0.17	0.08-0.19	0.07-0.09	0.46-0.53	0.51-0.60	20			
85 62 e 9 e 3(85.0 42 62.4 43 9.3 None	0.44-0.48 0.90-1.10	0.17–0.23 0.21–0.28	0.23-0.33	0.11–0.17								
62 e 9 e 3(62.4 43 9.3 None	0.90-1.10	0.21-0.28			0.08-0.24	0.05-0.07	0.47-0.50	0.50-0.61	20			
e 9 e 3(9.3 None			0.14-0.26	0.12-0.19								
e 9 e 3(9.3 None			0.14-0.26	0.12-0.19								
e 3(0.05-0.07	0.18-0.25			0.08-0.20	0.04-0.06	0.51-0.53	0.57-0.62	34			
e 3(0.05-0.07	0.18-0.25										
	30.0 5		direct drawn	0.20-0.32	0.06-0.15	0.09-0.19	0.04-0.05	0.48-0.51	0.62-0.65	25			
	30.0 5												
		0.05-0.07	0.19-0.25	0.22-0.32	0.06-0.14	0.09-0.20	0.05	0.51-0.53	0.50-0.58	16			
	40.0 5	0.03	0.24-0.27	0.22-0.29	0.09-0.10	0.22-0.23	0.03-0.04	0.52	0.60-0.61	2			
	40.0 5 51.1 9	0.03-0.07	0.24-0.27	0.15-0.31	0.09-0.10	0.02-0.23	0.03-0.04	0.52	0.62-0.66	34			
	67.3 19	0.01-0.06	0.19-0.27	0.21-0.32	0.05-0.15	0.09-0.23	0.02-0.04	0.52-0.56	0.52-0.62	23			
0.0	0-93.7 None	0.95 0.06	0.27-0.32	0.25-0.30	0.08-0.13	0.13-0.19	0.05-0.08	0.56-0.57	0.50-0.51	8			
	0–93.7 None 66.8 18	0.85-0.96 0.92-2.56	0.27-0.32	0.25-0.30	0.08-0.13	0.13-0.19	0.05-0.08	0.35-0.57	0.30-0.51	24	83-84	12	770-830
	76.1 32	0.55-1.21	0.19-0.36	0.16-0.27	0.09-0.15	0.06-0.20	0.07-0.08	0.50-0.52	0.50-0.60	21	*		
	1.1 2 20.0 10	0.84-1.16 1.34-1.77	0.18-0.38 0.37-0.38	0.24-0.30 0.25-0.26	0.11-0.15 0.09-0.13	0.08-0.15	0.06-0.07 0.06-0.07	0.46-0.51 0.52-0.53	0.45-0.49 0.57-0.58	5	* 88	5	600-640
obro 20	20.0 10	1.54-1.77	0.37-0.38	0.25-0.20	0.09-0.13	0.04-0.12	0.00-0.07	0.52-0.55	0.57-0.58	4	00	5	000-040
25	28.5 6	1.39-1.84	0.30-0.42	0.29-0.41	0.15-0.21	0.15-0.25	0.10-0.14	0.48-0.53	0.42-0.49	8	88-90	12	760-820
30	30.2 6	0.94-1.52	0.25-0.32	0.23-0.33	0.13-0.20	0.07-0.22	0.09-0.11	0.51-0.55	0.46-0.56	34	89-90	7	800-890
30	30.7 6	0.72-1.14	0.23-0.30	0.20-0.35	0.13-0.20	0.08-0.19	0.07-0.09	0.46-0.52	0.51-0.60	20	89-90	5	840-900
	05.0 10	0.47 0.47	0.00.0.00	0.00.0.00	011 017	0.00.0.21	0.05.0.06	0.40.0.40	0.50 0.(1		ale .		
8:	85.0 42	0.46-0.47	0.20-0.23	0.23-0.33	0.11-0.17	0.09-0.21	0.05-0.06	0.48-0.49	0.59-0.61	п	*		
6	62.4 43	0.91-1.11	0.21_0.28	0.14-0.26	0.12-0.19	0.08-0.20	0.04_0.06	0.50-0.53	0.50-0.62	33	90	5	860-880
0.	02.4 45	0.91-1.11	0.21-0.20	0.14-0.20	0.12-0.17	0.00-0.20	0.04-0.00	0.00-0.00	0.59-0.02	55	20	5	000-000
e 9	9.3 None	0.05-0.07	0.18-0.25	0.25-0.32	0.06-0.15	0.09-0.16	0.04-0.05	0.48-0.51	0.62-0.65	24	91	11	780-840
						1999-9999-9999-9999-9999-9999-9999-999		a sona constant.	utore - Adam	0.000	1000	100	11,902007585710
e 30	30.0 5	0.05-0.07	0.22-0.27	0.26-0.32	0.08-0.14	0.11-0.21	0.050	0.52-0.53	0.56-59	9	90-91	4	760
	40.0 5	0.03	0.24-0.27	0.22-0.29	0.09-0.10	0.22-0.23	0.030	0.52	0.60-0.61	2	91	14	760-770
													740-850 790-820
	5	30.7 6 85.0 42 62.4 43 9.3 None 30.0 5 40.0 5	30.7 6 0.72-1.14 85.0 42 0.46-0.47 62.4 43 0.91-1.11 9.3 None 0.05-0.07 30.0 5 0.05-0.07 40.0 5 0.03 51.1 9 0.03-0.07	30.7 6 0.72-1.14 0.23-0.30 85.0 42 0.46-0.47 0.20-0.23 62.4 43 0.91-1.11 0.21-0.28 9.3 None 0.05-0.07 0.18-0.25 30.0 5 0.05-0.07 0.22-0.27 40.0 5 0.03 0.24-0.27 51.1 9 0.03-0.07 0.19-0.24	30.7 6 0.72-1.14 0.23-0.30 0.20-0.35 85.0 42 0.46-0.47 0.20-0.23 0.23-0.33 62.4 43 0.91-1.11 0.21-0.28 0.14-0.26 9.3 None 0.05-0.07 0.18-0.25 0.25-0.32 30.0 5 0.05-0.07 0.22-0.27 0.26-0.32 40.0 5 0.03 0.24-0.27 0.22-0.29 51.1 9 0.03-0.07 0.19-0.24 0.22-0.31	30.7 6 0.72-1.14 0.23-0.30 0.20-0.35 0.13-0.20 85.0 42 0.46-0.47 0.20-0.23 0.23-0.33 0.11-0.17 62.4 43 0.91-1.11 0.21-0.28 0.14-0.26 0.12-0.19 9.3 None 0.05-0.07 0.18-0.25 0.25-0.32 0.06-0.15 30.0 5 0.05-0.07 0.22-0.27 0.26-0.32 0.08-0.14 1 9 0.03 0.24-0.27 0.22-0.29 0.09-0.10 5 0.03 0.24-0.27 0.22-0.31 0.06-0.15	30.7 6 0.72-1.14 0.23-0.30 0.20-0.35 0.13-0.20 0.08-0.19 85.0 42 0.46-0.47 0.20-0.23 0.23-0.33 0.11-0.17 0.09-0.21 62.4 43 0.91-1.11 0.21-0.28 0.14-0.26 0.12-0.19 0.08-0.20 9.3 None 0.05-0.07 0.18-0.25 0.25-0.32 0.06-0.15 0.09-0.16 30.0 5 0.05-0.07 0.22-0.27 0.26-0.32 0.08-0.14 0.11-0.21 40.0 5 0.03 0.24-0.27 0.22-0.29 0.09-0.10 0.22-0.23 51.1 9 0.03-0.07 0.19-0.24 0.22-0.31 0.09-0.15 0.09-0.20	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	30.7 6 0.72-1.14 0.23-0.30 0.20-0.35 0.13-0.20 0.08-0.19 0.07-0.09 0.46-0.52 85.0 42 0.46-0.47 0.20-0.23 0.23-0.33 0.11-0.17 0.09-0.21 0.05-0.06 0.48-0.49 62.4 43 0.91-1.11 0.21-0.28 0.14-0.26 0.12-0.19 0.08-0.20 0.04-0.06 0.50-0.53 9.3 None 0.05-0.07 0.18-0.25 0.25-0.32 0.06-0.15 0.09-0.16 0.04-0.05 0.48-0.51 30.0 5 0.05-0.07 0.22-0.27 0.26-0.32 0.08-0.14 0.11-0.21 0.050 0.52-0.53 40.0 5 0.03 0.24-0.27 0.22-0.29 0.09-0.10 0.22-0.23 0.030 0.52 51.1 9 0.03-0.07 0.19-0.24 0.22-0.31 0.09-0.10 0.22-0.23 0.030-0.04 0.51-0.52	30.7 6 0.72-1.14 0.23-0.30 0.20-0.35 0.13-0.20 0.08-0.19 0.07-0.09 0.46-0.52 0.51-0.60 85.0 42 0.46-0.47 0.20-0.23 0.23-0.33 0.11-0.17 0.09-0.21 0.05-0.06 0.48-0.49 0.59-0.61 62.4 43 0.91-1.11 0.21-0.28 0.14-0.26 0.12-0.19 0.08-0.20 0.04-0.06 0.50-0.53 0.59-0.62 9.3 None 0.05-0.07 0.18-0.25 0.25-0.32 0.06-0.15 0.09-0.16 0.04-0.05 0.48-0.51 0.62-0.65 30.0 5 0.05-0.07 0.22-0.27 0.26-0.32 0.08-0.14 0.11-0.21 0.050 0.52-0.53 0.56-59 40.0 5 0.03 0.24-0.27 0.22-0.29 0.09-0.10 0.22-0.23 0.030 0.52 0.60-0.61 51.1 9 0.03-0.07 0.19-0.24 0.22-0.29 0.09-0.10 0.02-0.23 0.030 0.52 0.60-0.61	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Notes: The summary of unzoned cores shows the composition of "fresh" spinel, with analyses removed that show evidence of post-crystallization metamorphic exchange (see text). Mg# = Mg/(Mg+Fe³⁺), Cr# = Cr/(Cr+Al+Fe³⁺), and Fe³⁺# = Fe³⁺/(Cr+Al+Fe³⁺). Fo = forsterite content. Olivine-spinel (OI-Sp) temperatures were calculated using a program provided by R. Sack, based upon the spinel solution model of Sack and Ghiorso (1991a, 1991b). * = no fresh olivine was found to analyze.

Table 2. Representative Cr-spinel analyses.

Rock: Hole:	G 895D	G 895D	OG 895D	OG 895E	OG 895E	OG 895E	OG 895E													
Core, section:	10W-1	10W-1	7R-2	8R-2	8R-2	8R-2	8R-2	1R-1	1R-1	2R-1	2R-1									
Interval (cm):	82-85	82-85	71-75	71-75	71-75	71-75	71-75	71-75	71-75	71-75	71-75	71-75	42-44	42-44	42-44	42-44	109-112	109-112	36-39	36-39
Analysis #:	B1-C	A1-C	A1-C	A1-R	4-IOC	5-IOR	11-IOC	16-C	D01-IPC	28-IPC	30-IPC	33-C	A1-C	B1-C	B1-R	D1-C	B1-C	A1-C	A1-IPC	A1-IP
SiO ₂	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.02
TiO,	0.88	0.95	2.36	2.14	2.05	1.88	2.49	2.22	2.26	2.14	1.45	0.93	1.06	1.18	0.90	0.57	0.84	1.16	1.50	1.16
Al ₂ O ₃	19.54	18.74	18.61	18.12	18.06	17.98	18.76	18.13	18.58	19.16	20.60	29.58	22.69	21.37	20.23	22.72	24.67	22.21	21.89	18.93
V_2O_3	0.32	0.31	0.59	0.56	0.61	0.57	0.52	0.48	0.54	0.46	0.38	0.17	0.27	0.36	0.30	0.22	0.18	0.36	0.37	0.32
Cr ₂ O ₃	44.47	43.81	37.97	38.60	38.15	37.20	35.45	36.99	37.37	38.12	37.19	29.26	40.59	39.98	39.95	40.77	36.91	39.08	41.30	40.31
FeO	22.76	23.92	31.24	32.00	33.23	34.88	32.64	31.56	31.36	30.72	28.45	27.42	21.13	25.27	29.95	21.51	25.44	26.22	21.49	33.87
MnO	0.25	0.28	0.45	0.49	0.45	0.55	0.40	0.40	0.42	0.38	0.36	0.28	0.20	0.24	0.47	0.19	0.24	0.24	0.25	1.11
NiO	0.08	0.11	0.14	0.11	0.14	0.09	0.14	0.09	0.10	0.13	0.04	0.10	0.15	0.11	0.09	0.09	0.15	0.12	0.13	0.08
ZnO	0.16	0.13	0.25	0.22	0.38	0.37	0.15	0.16	0.26	0.19	0.17	0.13	0.11	0.15	0.21	0.11	0.12	0.14	0.12	0.75
MgO	10.81	10.71	7.62	6.75	6.61	5.48	8.30	8.60	8.20	8.43	10.09	11.24	13.08	10.83	6.63	13.12	10.60	9.64	12.84	2.56
CaO	0.02	0.00	0.01	0.05	0.00	0.00	0.07	0.00	0.01	0.02	0.02	0.01	0.04	0.00	0.05	0.05	0.01	0.01	0.02	0.10
Total	99.29	98.96	99.24	99.04	99.69	99.00	98.92	98.64	99.10	99.75	98.75	99.12	99.32	99.49	98.78	99.35	99.17	99.18	99.91	99.21
RFeO	18.76	18.76	24.54	25.50	25.78	27.08	23.65	22.89	23.59	23.48	20.37	19.74	15.98	19.41	24.83	15.57	19.85	21.21	16.79	30.04
Fe ₂ O ₃	4.44	5.73	7.45	7.23	8.28	8.67	9.99	9.64	8.63	8.04	8.98	8.54	5.72	6.52	5.69	6.61	6.21	5.56	5.23	4.25
Total	99.73	99.53	99.99	99.76	100.52	99.87	99.92	99.60	99.97	100.56	99.65	99.98	99.89	100.14	99.35	100.01	99.79	99.74	100.43	99.64
Fe#	0.054	0.071	0.097	0.095	0.108	0.114	0.130	0.125	0.112	0.103	0.112	0.100	0.068	0.079	0.072	0.078	0.074	0.068	0.063	0.056
Cr#	0.571	0.567	0.521	0.532	0.523	0.515	0.486	0.505	0.510	0.513	0.486	0.359	0.508	0.512	0.529	0.504	0.464	0.504	0.523	0.555
Mg#	0.507	0.504	0.356	0.321	0.314	0.265	0.385	0.401	0.382	0.390	0.469	0.504	0.593	0.499	0.322	0.600	0.488	0.447	0.577	0.132

Notes: Analyses (electron microprobe, by J. Allan) are recalculated by stoichiometry according to the methods of Carmichael (1967). RFeO = recalculated FeO. Mg#, Cr#, and Fe³⁺# as in Table 1. C = core, R = rim, 1 = inclusion, with O and P denoting inclusions in olivine and plagioclase, respectively. Sample numbers preceded by a letter and a number (e.g., A1) are keyed to analytical transects shown in Figure 6; for example, analyses A1-C and A1-R from Sample 147-895D-7R-2, 71–75 cm, refer to the core and rim analyses of transect A1 in the figure.

Rock: Hole: Core, section: Interval (cm): Analysis #;	OG 895E 2R-1 36–39 B1-IPC	OG 895E 2R-1 36–39 B1-IPR	OG 895E 2R-1 36–39 12	T 895C 4R-1 63–68 1-C	T 895C 4R-1 63-68 5-C	T 895C 4R-1 63-68 7-C	T 895C 4R-1 63-68 8-C	T 895C 4R-2 75–78 C1-C	T 895C 4R-2 75–78 C2-C	T 895C 4R-2 75–78 B1-C	T 895C 4R-2 75–78 B2-C	T 895C 4R-2 75–78 A1-C	T 895C 4R-2 75–78 A2-R	T 895C 4R-2 126-128 AA1-C	T 895C 4R-2 126–128 BB1-C	T 895C 4R-2 126–128 BB2-C	D 895D 9R-1 101-105 A1-C	D 895D 9R-1 101-105 A2-C	D 895D 9R-1 101–10 A2-R
SiO ₁	0.06	0.00	0.00	0.00	0.00	0.02	0.00	0.00	0.05	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	1.77	1.40	1.32	1.44	1.65	1.61	1.84	0.95	1.04	1.35	1.50	1.42	1.35	1.10	0.83	0.74	0.00	0.45	0.44
Al ₂ O ₃	21.05	25.47	14.82	19.19	18.65	20.00	16.29	21.58	20.81	18.17	18.04	20.17	22.05	20.57	22.98	23.75	25.41	25.66	24.61
V ₂ O ₃	0.38	0.40	1.01	0.34	0.36	0.42	0.37	0.27	0.28	0.31	0.32	0.29	0.28	0.25	0.25	0.23	0.21	0.21	0.21
Cr.O.	41.26	34.15	40.18	40.68	40.00	38.47	40.93	40.23	40.45	41.50	41.44	40.79	36.99	40.44	39.96	39.61	39.55	40.17	39.96
FeO	21.85	30.86	35.22	26.92	28.23	28.46	30.80	23.67	24.41	27.25	26.70	24.73	28.86	24.75	22.33	22.00	20.61	19.40	21.87
MnO	0.26	0.72	0.71	0.32	0.36	0.32	0.36	0.24	0.24	0.33	0.29	0.28	0.37	0.35	0.25	0.27	0.30	0.28	0.31
NiO	0.09	0.10	0.05	0.15	0.20	0.21	0.17	0.14	0.14	0.16	0.16	0.19	0.15	0.16	0.14	0.17	0.14	0.13	0.12
ZnO	0.10	0.23	0.25	0.16	0.24	0.19	0.15	0.12	0.12	0.22	0.10	0.11	0.17	0.15	0.12	0.18	0.16	0.21	0.19
MgO	12.84	6.38	4.71	10.60	9.94	10.47	8.89	12.12	11.83	9.97	10.65	11.49	8.52	11.16	13.00	13.17	13.01	13.42	11.27
CaO	0.00	0.03	0.00	0.00	0.00	0.00	0.01	0.01	0.00	0.01	0.00	0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00
Total	99.66	99.74	98.27	99.80	99.63	100.16	99.80	99.33	99.36	99.28	99.20	99.47	98.75	98.95	99.85	100.12	99.86	99.93	98.98
RFeO	16.97	26.53	27.02	19.64	20.55	20.20	22.13	17.20	17.70	20.10	19.32	18.34	22.77	18.37	16.13	15.89	16.06	15.45	18.25
Fe ₂ O ₁	5.43	4.81	9.12	8.09	8.53	9.18	9.63	7.19	7.46	7.94	8.21	7.10	6.77	7.10	6.89	6.78	5.06	4.39	4.02
Total	100.20	100.22	99.19	100.61	100.49	101.08	100.77	100.05	100.11	100.08	100.02	100.19	99.43	99.66	100.54	100.80	100.37	100.37	99.38
Fe#	0.066	0.060	0.122	0.100	0.107	0.113	0.123	0.086	0.090	0.099	0.103	0.087	0.084	0.087	0.081	0.079	0.059	0.051	0.048
Cr#	0.530	0.445	0.566	0.528	0.527	0.499	0.550	0.508	0.515	0.545	0.544	0.525	0.485	0.519	0.495	0.486	0.481	0.486	0.496
Mg#	0.574	0.300	0.237	0.490	0.463	0.480	0.417	0.557	0.544	0.469	0.496	0.528	0.400	0.520	0.589	0.596	0.591	0.608	0.524

Table 2 (continued).

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28-1 4R-1 2R-1 4R-1 7-10 10-12							711	711	711	711	711	
			95C 895C	895C 895D	D 895D	895D	895D	895D	895D	895D	895D	
						5R-1	5R-1	5R-1	7R-2	7R-2	7R-2	
			~		~	83-86	83-86	83-86	123-126	123-126	123-126 1	
			2			BI-C	A1-C	A2-C	AI-C	B1-C	B1-R	CI-C 5
0.00 0.00						0.00	0.02	0.00	0.00	0.00		
						0.05	0.05	0.05	0.05	0.05		
						25.08	25.25	25.42	23.88	25.01		
						0.22	0.24	0.22	0.22	0.23		
						43.19	43.23	43.30	44.25	43.85		
						17.16	16.78	16.73	17.91	17.60		
						0.25	0.25	0.20	0.25	0.24		
0.12		0.12				0.08	0.12	0.10	0.13	0.12		
0.15		0.15				0.16	0.15	0.11	0.12	0.14		
12.04	-	12.04				13.61	13.84	13.93	13.21	13.32		
0.00		0.00				0.00	0.00	0.00	0.01	0.00		
99.53	6	99.53		99.95 99.72	2 99.75	08.66	99.93	100.07	100.02	100.56	100.50	99.68 99.99
						14.76	14.47	14.48	15.25	15.43		
						2.67	2.56	2.50	2.96	2.42		
76.66	6	76.99	.97 100.46	-	_	100.07	100.18	100.32	100.32	100.80		-
0	0	0		-		0.031	0.029	0.028	0.034	0.028	0.034	0.035
			0.517 0.517	0.517 0.515	15 0.508	0.520	0.519	0.518	0.535	0.525	0.522	0.530
						0.622	0.630	0.632	0.607	0.606	0.618	0.620

increases in Cr# (Allan et al., 1988), as Fe2+ and Cr3+ and Mg2+ and Al3+ are energetically coupled in the spinel structure. This relationship is illustrated by the cores of spinel grains from olivine gabbro Sample 147-895D-7R-2, 71-75 cm, and troctolite Samples 147-895C-4R-1,63-68 cm, 147-895C-4R-2, 75-78 cm, and 147-895C-4R-2, 126-128 cm, as shown in Table 2. Similar changes in spinel chemistry, where altered grains or rims of grains show drops in Mg# without change in Cr#, have been reported in a wide range of metamorphosed rocks, including both low-temperature (<200°C) hydrothermally altered spinel in basalt (Allan, 1992) and in hightemperature granulite facies rocks (Sack, 1980). Engi (1983) gave numerous references that show a relatively low blocking temperature for Mg-Fe2+ exchange between olivine and spinel; these studies and the Hess Deep spinel disequilibrium zoning patterns show that kinetic barriers for trivalent cation exchange are higher than for divalent cation exchange in spinel, leading to more sluggish exchange (Roeder et al., 1979; Engi and Evans, 1980; Sack, 1980). Apparently, spinel Ti4+ is more mobile than the trivalent cations under these metamorphic conditions as well.

Within individual samples, alteration of spinel margins was erratic, occurring in some grains and not in others. Margins of spinel inclusions were often subjected to metamorphic alteration, as illustrated by spinels from Sample 147-895E-2R-1, 36–39 cm, (transects of grains shown in Fig. 6; these occur within An₈₅₋₈₇ plagioclase) and by analyses 4-IOC and 5-IOR from Sample 147- 895D-7R-2, 71–75 cm, a core-rim pair of a $80 \times 100 \,\mu$ m inclusion in olivine. These spinels show evidence of loss in Ti and a lowering in Mg#. Scowan et al. (1991) found that spinel inclusions in olivine from the Kilauea Iki lava lake became richer in TiO₂ as the lava lake crystallized, indicating diffusive movement of Ti through olivine to the spinel; an opposite movement of Ti appears to have occurred here. Fractures in the host olivines and plagioclase, some lined with secondary clays, may have provided pathways for metamorphic fluids.

Chemistry of "Fresher" Spinel

To more closely approximate original Site 895 spinel compositional relationships, analyses considered significantly altered during the lower temperature metamorphic overprints were removed from consideration. Analyses within individual grains were considered acceptable for retention only if two or more analyses in a transect had similar Mg# values, a value judgement based on the fact that the cores of all grains analyzed on transects had little zoning in Mg# (e.g., Fig. 6)any analyses with significantly lower Mg# than in these transects were excluded. Comparison with other analyses showed that the cores of some spinel grains smaller than 100 µm were relatively low in Mg# for a given TiO2 content (as shown in Table 2 banalyses 4-IOC and 5-IOR from Sample 147-895D-7R-2, 71-75 cm, in comparison to other analyses from this sample); these could reflect extensive higher temperature, metamorphic olivine-spinel Mg-Fe2+ exchange (Ozawa, 1983) prior to lower amphibolite metamorphism, and they were excluded as well. The upper part of Table 1 and Figure 7 show the "uncorrected" data set, whereas the lower part of Table 1 and Figure 8 show the smaller, "fresher" data set.

After removal from consideration of analyses strongly affected by lower temperature metamorphic exchange, the crystal chemical variations within the Site 895 spinels become clearer and more systematic, both within individual samples and for the entire sample group (compare Figs. 7 and 8). Within the harzburgite samples, spinel composition is typical of those from abyssal, depleted harzburgite (Dick, 1989; Dick and Fisher, 1984; Dick and Bullen, 1984), with high Cr#, moderate Mg#, low V₂O₃ and ZnO, and very low TiO₂ (typically $\leq 0.05\%$) and Fe³⁺/(Cr + Al + Fe³⁺) (Fe³⁺#; ≤ 0.05) (Tables 1, 2; Figs. 7, 8). Internal zoning, other than the somewhat variable V₂O₃ and ZnO, is limited within individual crystals, but variations exist grainto-grain within a single sample as well as between samples (Table 2;

Table 2 (continued).

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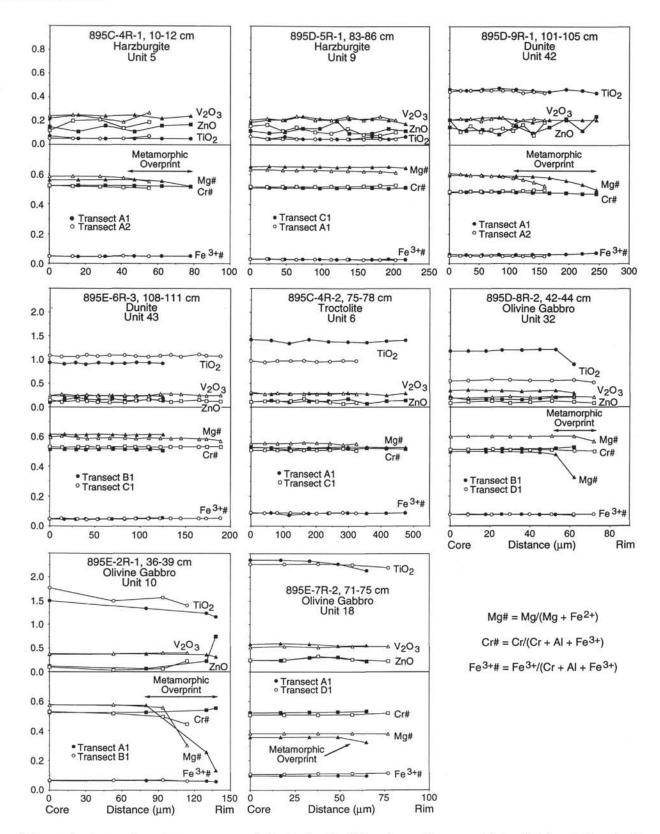


Figure 6. Core-to-rim electron microprobe transects across spinel grains from Site 895 samples, avoiding areas of obvious ferritchromite formation. Transect pairs are analyzed from grains within individual thin sections; representative analyses from these transects are given in Table 2. Post-crystallization metamorphic exchange has caused Mg# or Mg# and TiO₂ to drop toward the rims of some transects; ZnO rises in extreme cases of alteration (transects of Sample 147-895E-2R-1, 36–39 cm). Note that spinel TiO₂ in the dunitic samples varies strongly from sample to sample, and that troctolites and olivine gabbros may show strong variance in TiO₂ and Mg# in spinels within a single thin section (Samples 147-895C-4R-2, 75–78 cm, and 147-895D-8R-2, 42–44 cm).

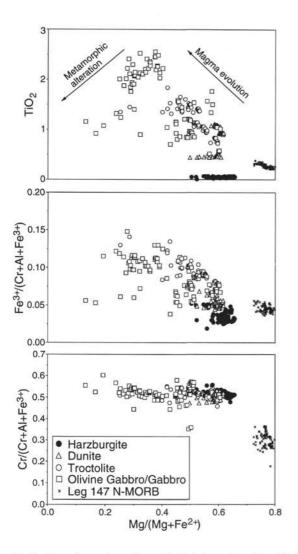


Figure 7. Electron microprobe analyses of spinels from the various Site 895 rock types. Also shown for comparison are Cr-spinels analyzed from basaltic lava and dike samples from Site 894 that have an N-MORB compositional signal (Allan et al., this volume). Also shown are the inferred trends in TiO_2 and Mg/(Mg + Fe²⁺) for the troctolite and gabbro spinels from processes associated with magma evolution and post-crystallization metamorphic exchange (see text).

Fig. 6). In addition, analyses of vermicular spinel (analysis 5, Sample 147-895D-7R-2, 123–126 cm, and analysis 5, Sample 147-895D-4R-1, 34–37 cm, in Table 2) are little different than for other, larger subangular or subrounded spinels, other than having slightly higher ZnO.

Dunitic spinels show substantial differences in composition compared with those in harzburgite. Although morphologically similar and overlapping in Cr#, they contain 10 to 20 times the amount of TiO₂, with slightly higher average Fe³⁺# and NiO as well (Tables 1, 2; Figs. 6, 8). Considering the two dunite samples as a group, there is a positive correlation between TiO₂ and Cr# (also found in other Site 895 dunites by Dick and Natland, this volume; see their fig. 10). A slight negative correlation between TiO₂ and Mg# is visible within dunite Sample 147-895E-6R-3, 108–111 cm. Like the spinels in harzburgite, the dunite spinels show only minor internal zoning and exhibit minor grain-to-grain variation within a single sample. Nevertheless, the two dunite samples analyzed have substantial differences in spinel TiO₂ content, with the spinels in Sample 147-895E-6R-3, 108–111 cm, having roughly twice the amount of TiO₂ of those from

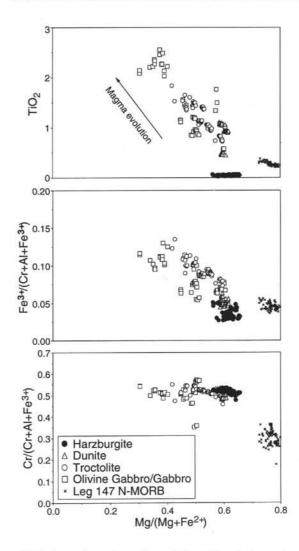


Figure 8. Electron microprobe analyses of "fresh" Cr-spinel cores from the various Site 895 rock types, omitting spinels affected by elemental exchange associated with metamorphic alteration and thereby giving a clearer view of the original Cr-spinel crystal chemistry.

Sample 147-895D-9R-1, 101–105 cm. Additional data reported by Dick and Natland (this volume) extend the range of Site 895 dunitic spinel TiO_2 from 0.5% to 2.0%. Of interest is that Sample 147-895E-6R-3, 108–11 cm, (with Ti-rich spinel) occurs adjacent to a unit of troctolite, whereas Sample 147-895D-9R-1, 101–105 cm, (with relatively Ti-poor spinel) occurs between units of harzburgite and is not adjacent to gabbro or troctolite.

Within the gabbroic and troctolitic spinels, no systematic difference was noted in individual samples in the chemistry of spinels that occur as intergranular grains compared to those that occur as inclusions within olivine or plagioclase (Table 2). Systematic variations for the group are well illustrated in Figure 8, showing strong negative correlations of TiO₂, Fe³⁺#, and Cr# with Mg#, with the latter correlation illustrating the crystal chemical effects of coupled Fe²⁺–Cr³⁺ and Mg²⁺–Al³⁺ substitution within spinel discussed previously. The negative correlation between Cr/(Cr + Al) with Mg#, shown by Dick and Natland (this volume), is especially strong. Gabbro, olivine gabbro, and troctolitic spinels are strikingly enriched in TiO₂ and broadly enriched in V₂O₃ over harzburgitic spinel, but show little overall difference in NiO and ZnO contents (Table 1; Fig. 8). Overall, the range in Mg# exhibited by the gabbro and olivine gabbro spinels is substantial, overlapping in Mg# with the dunitic and troctolitic spinels, but with many having a Mg# of <0.4 (Fig. 8). The presence of spinel with low Mg# is consistent with the relatively low Fo content of the associated olivine in both troctolitic and gabbroic samples, particularly in Sample 147-895D-7R-2, 71–75 cm (Table 1).

Sample 147-895D-7R-2, 71–75 cm, contains a broad range of spinel compositions, with the large number of grains analyzed (16) allowing a careful examination of spinel compositional systematics within a single thin-section sample of olivine gabbro. In considering the relatively unaltered cores, strong positive correlations of TiO_2 exist with V_2O_3 and MnO, with weaker positive correlations existing with TiO_2 and ZnO and Cr#. In contrast, there is a strong negative correlation between spinel Mg# and TiO₂, and no correlations occur between TiO_2 and Fe^{3+#} and NiO. These relationships are illustrated in Table 2 by analyses A1-C, 11-C, 16-C, D1-C, 30-C, and 33-C. Spinels from other gabbro samples share these relationships, with Sample 147-895D-8R-2, 42–44 cm, also showing a broad range in spinel composition (Table 1; analyses A1-C, B1-C, and D1-C in Table 2).

The Hole 895C troctolite samples most likely originate from the same intrusive event owing to their proximity and nearly identical physical character. They show poorer but similar correlations within their Cr-spinel systematics when considered as a group. Tables 1 and 2 and Figure 6 show results of nine transects and eight other grains analyzed from Samples 147-895C-4R-2, 75-78 cm, 147-895C-4R-2, 126-128 cm, and 147-895C-4R-1, 63-68 cm, with all showing the same curious lack of zoning within the large grains, some of which re skeletal. Plate 1F shows a photomicrograph of the crystal of electron microprobe transect C1 from Sample 147-4R-2, 75-78 cm, (Fig. 6) which contains inclusions of irregular plagioclase (An₈₄₋₈₆) and clinopyroxene (En48-49, Fs0.04-0.05, Wo0.46-0.47). Table 4 gives representative analyses of these grains, for which the unexsolved nature, small compositional range, and apparent protection from lower amphibolite to zeolite metamorphic alteration indicate that their compositions may approximate magmatic values.

A striking feature of both the troctolites and the gabbro samples is that fresh cores of individual spinel grains show little or no zoning, yet the differences in TiO₂ between individual grains within a thin section can be quite significant. For example, the two electron microprobe transects shown for the troctolite Sample 147-895C-4R-2, 75– 78 cm, in Figure 6 differ by nearly 0.5% in TiO₂, yet these two 0.5cm-sized grains are only about a centimeter apart in thin section. A similar distance separates the two spinel transects shown in Figure 6 from Sample 147-895D-8R-2, 42–44 cm. These grains show more than a two-fold variation in TiO₂ despite sharing similar Mg#.

DISCUSSION

Gabbroic Spinel Ti and Fe3+ Enrichment

The spinels in gabbro and troctolite exhibit a large range in composition, with the dominant form of variation shown by the increase in Ti and Fe3+# with declining Mg# (Table 1; Fig. 8). At low Mg#, TiO2 contents commonly exceed 2%, similar to that found in other Cr-spinels from abyssal and ophiolitic plagioclase-rich troctolites and plagioclase dunites (Cannat et al., 1990; Hekinian et al., 1993; Girardeau and Francheteau, 1993; Dick, 1989; Hebert et al., 1983). These TiO₂ (and Fe^{3+#}) values are much higher than those observed in N-MORB Cr-spinel (Fig. 8; Allan et al., this volume; Allan et al., 1988) and are similar in enrichment to those found in oceanic alkali basalt (Allan et al., 1993, 1994; Batiza and Vanko, 1984). Nevertheless, numerous pieces of evidence require that the parental basaltic melts to the gabbroic Cr-spinels be of an N-MORB compositional character. This evidence includes (1) bulk composition of Site 895 gabbroic rocks, which are depleted in incompatible elements (Gillis, Mével, Allan, et al., 1993; J. Allan, unpub. data); (2) electron and ion microprobe analysis of gabbroic clinopyroxene (Dick and Natland, this volume; H.J.B. Dick and J. Natland, unpub. data), which shows

compositions depleted in Ti and the rare earth elements; (3) low Ti contents of the clinopyroxene inclusions in skeletal, troctolite spinel, with compositions typical or lower than that of N-MORB clinopyroxene (e.g., Allan et al., 1987; Perfit and Fornari, 1983; Mazzullo and Bence, 1976); and (4) K-poor, An-rich contents of the plagioclases analyzed, especially considering the inclusions in troctolitic spinel.

Given that the impregnating melt is of N-MORB character, the strong negative correlation of Ti and Fe3+# with Mg# most likely reflects spinel crystallizing from or reequilibrating with melt that is evolving as a result of crystallization and melt-wall rock interaction. Both processes will increase melt Ti and Fe3+ as long as the magma mass is decreasing, because these elements are excluded from plagioclase and olivine. The large variations in spinel TiO₂ within a single thin section argue that these variations are related to spinel reequilibrating with increasingly small pockets of intercrystalline, evolving melt during cooling after or during melt impregnation. The presence of evolved melt is supported by the relatively fayalitic nature of the adjacent olivines in the gabbros, particularly within Sample 147-895D-7R-2, 71-75 cm (Tables 1 and 3). Within other samples, substantial variation in spinel Ti with little variation in spinel Mg# (such as shown by transects B1 and D1 of Sample 147-895D-8R-2, 42-44 cm, in Fig. 6) may reflect varying degrees of wall-rock interaction, with the Mg# buffered by the surrounding peridotite as the volume of melt decreased during crystallization.

Implications of Spinel in Dunite

The composition of the spinels in dunite exerts strong constraints on models of dunite origin. The high TiO2 of the dunite spinel, approximately 10 times that of Hess Deep harzburgite spinel, rules out the possibility that the dunites analyzed here represent residual peridotite that has been melted to a pyroxene-out condition, as Ti is incompatible during mantle partial melting (e.g., Dick, 1989). In addition, the Cr# of dunite spinel is similar to that of harzburgite spinel (Table 1; Fig. 8), whereas it should be higher if the dunites represent residua from greater partial melting (Dick and Bullen, 1984). The lack of cumulate mineral textures, the similarly high Mg# of the dunite and harzburgite tectonite olivines, and the typical stratigraphic occurrence of dunite both above and below gabbro are incompatible with the Hess Deep dunites originating as simple cumulates from a magma chamber, as inferred for similar loose dunite and troctolite dredged from the Garrett Fracture Zone (Hebert et al., 1983). Instead, the high Ti contents of the dunite spinels, and the proximity of dunite to melt impregnations, indicate that the dunites formed from meltwall rock interaction, with a net outflow of Ti into the wall rock from the melt (Cannat et al., 1990). The high Ti contents of the dunite spinels are similar to those found within some ophiolitic podiform chromite bodies; these bodies are enclosed within dunitic envelopes that are themselves surrounded by harzburgite with Ti-poor spinel and are proposed to result from melt-wall rock interaction surrounding channels of upward melt flow (Dick, 1977, 1989; Arai and Yurimoto, in press; Leblanc and Ceuleneer, 1992; Lago et al., 1982).

The impregnated Hess Deep peridotites clearly illustrate processes of melt-wall rock disequilibrium, indicating that the melts could not have been generated from a similar peridotite in the immediate vicinity. The easiest explanation for the origin of this disequilibrium is to propose that the invading melt originated at a pressure substantially greater than that during melt impregnation. Melt originating from a given peridotite composition will have lower a_{SiO_2} at greater pressures of partial melting (e.g., Jacques and Green, 1980; Basaltic Volcanism Study Project, 1981); therefore, melt generated at higher pressure that is percolating through similar peridotite at substantially lower pressure will necessarily be out of equilibrium with the surrounding wall rock. The result during melt-wall rock interaction will thermodynamically drive dissolution of relatively Si-rich phases, such as orthopyroxene and clinopyroxene, and drive precipitation of relatively Si-poor phases, such as olivine and spinel (Kelemen, 1990; Nicolas, 1989; Fisk, 1986; Dick, 1977; Sinton, 1977), in order to inTable 3. Representative olivine analyses.

Rock:	OG	OG	OG	OG	T	T	T	T	T	T	T	T	T	
Hole:	895E	895E	895D	895D	895C	895C	895C	895C	895C	895C	895C	895C	895C	
Core, section:	2R-1	2R-1	7R-2	7R-2	4R-1	4R-1	4R-1	4R-2	4R-2	4R-2	4R-2	4R-2	4R-2	
Interval (cm):	36–39	36–39	71–75	71-75	63–68	63-68	63–68	75–58	75–58	75–58	75–58	75–58	126–128	
Analysis:	2	4	2	3	1	3	7	1	4	5	6	7	8	
SiO ₂	40.29	40.17	39.29	39.40	40.32	40.32	40.48	40.25	40.04	40.38	40.44	40.44	40.27	
FeO	11.59	11.43	15.08	16.03	10.39	10.79	10.89	9.86	10.84	10.16	10.43	10.56	10.41	
MnO	0.18	0.13	0.21	0.28	0.18	0.17	0.15	0.14	0.20	0.15	0.18	0.18	0.16	
NiO	0.26	0.26	0.22	0.22	0.30	0.27	0.30	0.31	0.29	0.29	0.29	0.32	0.33	
MgO	47.79	47.97	45.00	44.34	49.00	48.49	48.26	48.99	48.11	48.38	48.77	48.33	48.36	
CaO	0.00	0.03	0.03	0.03	0.04	0.05	0.06	0.03	0.05	0.07	0.03	0.05	0.05	
Total	100.11	100.00	99.83	100.29	100.23	100.08	100.13	99.58	99.52	99.42	100.13	99.88	99.58	
Fo	0.880	0.882	0.842	0.831	0.894	0.889	0.888	0.899	0.888	0.895	0.893	0.891	0.892	
Near Cr-spinel	12	0	4	2, 3	1	5	8	A1	B2	B1	C1	C2	AA1	
Rock:	T	T	D	D	HZ	HZ	HZ	HZ	HZ	HZ	HZ	HZ	HZ	HZ
Hole:	895C	895C	895E	895E	895A	895A	895C	895D	895D	895D	895D	895D	895D	895D
Core, section:	4R-2	4R-2	6R-3	6R-3	2R-1	2R-1	4R-1	2R-1	4R-1	5R-1	5R-1	7R-2	7R-2	7R-2
Interval (cm):	126–128	126–128	108–111	108–111	7-10	7–10	10–12	123–126	34–37	83-86	83–86	123–126	123–126	123–120
Analysis:	10	11	3	4	2	7	2	1	12	3	7	6	9	11
SiO ₂	40.62	40.54	40.73	40.87	40.21	40.76	40.54	40.15	40.38	41.05	41.06	40.18	40.71	40.39
FeO	10.39	10.15	9.88	9.47	8.88	8.51	8.86	9.01	8.91	8.17	8.63	8.88	8.94	9.09
MnO	0.13	0.16	0.16	0.14	0.10	0.13	0.10	0.12	0.11	0.12	0.15	0.11	0.16	0.14
NiO	0.35	0.32	0.38	0.40	0.40	0.43	0.41	0.44	0.35	0.39	0.36	0.44	0.40	0.41
MgO	48.51	48.63	50.46	50.54	50.21	50.30	49.94	49.60	49.75	50.45	50.00	49.87	50.29	50.02
CaO	0.07	0.05	0.15	0.18	0.02	0.06	0.03	0.06	0.05	0.08	0.03	0.04	0.06	0.01
Total	100.06	99.84	101.76	101.60	99.82	100.18	99.89	99.38	99.54	100.25	100.22	99.53	100.55	100.05
Fo	0.893	0.895	0.901	0.905	0.910	0.913	0.909	0.907	0.909	0.917	0.912	0.909	0.909	0.90°
Near Cr-spinel	BB2	BB1	B1	C1	A1	C1	A2	A1	5	B	C	B1	C1	5

Notes: Electron microprobe analyst, J. Allan. OG = olivine gabbro; T = troctolite; D = dunite; HZ = harzburgite; Fo = olivine forsterite content. Labels of adjacent analyzed spinel grains are also given.

crease the a_{SiO_2} of the ascending melt. Recent geochemical modeling (Kelemen et al., in press) indicates that the wall rock may in fact be entirely replaced during this process, with the precipitation of large amounts of olivine. Therefore, dunitic pods containing spinel within mantle harzburgite may be relicts of melt-upflow channels, even where no obvious melt impregnations are present. Bodies of isolated dunite, sampled during Leg 147 without direct evidence of adjacent impregnated melt (such as the Hole 895D Unit 42 dunite of Sample 147-895D-9R-1, 101–105 cm), may have shared such an origin.

Other recent work regarding the origin of abundant Cr-spinel in podiform chromite bodies suggests that magma mixing, involving Sirich secondary melt generated during melt–wall rock interaction, may be involved in the rapid precipitation of Cr-spinel during basaltic melt–peridotite wall rock interaction (Arai and Yurimoto, in press). The skeletal troctolite spinels from Hole 895C, apparently rapidly grown, probably reflect crystallization by such a process; their widely varying Ti contents provide evidence for a complex series of melts that vary in composition because of fractionation and disequilibrium exchange on a centimeter-sized scale (Cannat et al., 1990; Dick and Kelemen, 1991; Leblanc and Ceuleneer, 1992; Kelemen et al., in press).

Constraints on Conditions and Location of Impregnation

The nature of occurrence, relative stratigraphy, and composition of the Site 895 spinel impose constraints on the physical conditions and possible location of melt impregnation. The conditions must be such to allow dispersal of exotic melt away from melt channels into intergranular networks in the surrounding peridotite, thereby forming dunite with troctolite, which sometimes was preserved on the margins of the melt channels. The relatively evolved nature and extensive recovery of some of the gabbroic rocks (more than 1 m of continuous core for Units 2 and 10, Hole 895E, and more than 0.5 m for Unit 28, Hole 895E; Gillis, Mével, Allan, et al., 1993) indicate that much of

the melt must have been channeled through dikelike bodies. The ambient temperature of the wall rock must have been low enough to preserve evidence of obvious disequilibria within the gabbroic and troctolitic rocks, as represented by individual samples with strongly variable spinel Ti contents. Therefore, during impregnation, the peridotite was (1) within the plagioclase stability field; (2) substantially shallower than typical depths associated with MORB genesis (e.g., significantly shallower than about 8-9 kbar pressure or about 25 km depth; Walter and Presnall, 1994; Presnall and Hoover, 1987; Kinzler and Grove, 1992a, 1992b; Nui and Batiza, 1991; McKenzie and Bickle, 1988; Klein and Langmuir, 1987) in order to account for melt-wall rock disequilibria; (3) no warmer than the liquidus of the impregnating melt (to preserve spinel compositional disequilibria); and (4) warm enough to allow melt impregnation by ductile deformation (Kennedy et al., this volume, found no evidence for microstructural olivine deformation in the dunites, requiring emplacement temperatures of at least 800°C). In addition, melt impregnation occurred most probably near the base of the crust, in order to allow uplift and exposure on the seafloor at a point that is relatively close to the spreading ridge.

Olivine-spinel geothermometry (Sack and Ghiorso, 1991a, 1991b) provides minimum melt impregnation temperatures (Table 1). These temperatures, derived from adjacent olivine-spinel pairs, represent metamorphic cooling reequilibration temperatures. To minimize compositional variation caused by post-crystallization exchange, the cores of adjacent spinel-olivine pairs were used. Sample 147-895E-2R-1, 36–39 cm, an olivine gabbro, gives the lowest temperatures, which most likely reflect spinel reequilibration during hydrothermal alteration that produced the Zn-rich spinel margins. Otherwise, maximum temperatures of about 880°–900°C were found for the troctolites and dunites, with somewhat lower maximum temperatures given by the olivine gabbros and harzburgites. These temperatures represent minimum estimates for ambient wall-rock temperature during impregnation, and almost certainly reflect some

Rock: Hole:	OG 895D	OG	OG	OG 895E	OG 895E	T 895C	T 895C
		895D	895D	2R-1		4R-2	
Core, section:	7R-2	7R-2	7R-2		2R-1		4R-2
Interval (cm):	71-75	71-75	71-75	36-39	36-39	75-78	75-78
Analysis #:	2	3	4	1	3	21	4I
Plagioclase							
SiO ₂	48.13	46.30	46.97	46.62	46.53	47.00	46.47
Al ₂ Õ ₃	33.45	34.69	34.35	34.48	34.73	34.49	34.82
FeO	0.17	0.15	0.17	0.27	0.29	0.19	0.16
MgO	0.02	0.02	0.02	0.16	0.09	0.03	0.03
CaO	15.93	17.64	16.78	17.69	17.58	17.00	17.52
Na ₂ O	2.49	1.61	1.91	1.48	1.65	1.74	1.59
K ₂ Ô	0.00	0.02	0.01	0.00	0.02	0.01	0.01
Total	100.19	100.42	100.20	100.70	100.87	100.46	100.59
An	77.9	85.7	82.9	86.9	85.4	84.3	85.9

Table 4. Selected plagioclase and clinopyroxene analyses.

Rock:	Т	Т	Т
Hole:	895C	895C	895C
Section:	4R2	4R2	4R2
Interval (cm):	75-78	75-78	75-78
Analysis #:	51	61	71
Clinopyroxene			
SiO ₂	51.27	52.09	51.42
TiO ₂	0.72	0.69	0.69
Al ₂ Ô ₃	3.10	2.60	2.97
Cr ₂ O ₃	1.23	0.92	0.93
FeO	3.29	2.66	3.02
MnO	0.07	0.08	0.09
NiO	0.08	0.02	0.06
MgO	16.98	17.15	16.75
CaO	22.45	23.28	22.99
Na ₂ O	0.37	0.35	0.35
K ₂ Ŏ	0.00	0.00	0.01
Total	99.56	99.84	99.28
Mg#	0.902	0.920	0.908
En	0.486	0.485	0.479
Fs	0.053	0.042	0.048
Wo	0.462	0.473	0.473

Notes: Electron microprobe analyst, J. Allan. OG = olivine gabbro, T = troctolite; An = anorthite content; $Mg\# = Mg/(Mg + Fe^{2+})$; En = enstatite content; Fs = ferrosilite content; Wo = wollastonite content.

amounts of spinel-olivine Mg-Fe²⁺ exchange during extended cooling.

The temperature of the wall rock during melt impregnation may also be estimated by approximating magmatic temperatures of MORB-like melts in equilibrium with the most Fe-rich olivine compositions in the gabbroic melt impregnations (unzoned olivine grains of Fo₈₃₋₈₄ composition in Sample 147-895D-7R-2, 71-75 cm) at appropriate pressures. These impregnating melts would have rapidly solidified if they had cooled below a point where they were in equilibrium with the olivine phase; indeed, the impregnating melts may be envisioned as existing within a crucible made up mostly of olivine with subordinate orthopyroxene. The experimental study by Bender et al. (1978), in which phase relations and compositions of a primitive FAMOUS basalt from 1 bar to 15 kbar were examined, provides a basis for approximating the temperature of the melts in equilibrium with olivine of Fo83-84 composition in the last stages of cooling, and therefore an estimate of the wall-rock temperature during melt impregnation. The FAMOUS basalt used by Bender et al. (1978) is an appropriate choice for comparison, as both it and the Leg 147 Hess Deep gabbroic rocks are relatively depleted in incompatible elements and are interpreted to have been the products of large amounts of partial melting from depleted sources (Dick and Natland, this volume). Interpolating between the 1 bar and 6 kbar Bender et al. (1978) olivine compositional data and comparing those data with the olivine data presented here leads to wall-rock estimates of about 1230°C at 3 to 5 kbar model pressures for the melt impregnation represented by Sample 147-895D-7R-2, 71-75 cm.

This temperature estimate may have been affected by post-crystallization Mg-Fe²⁺ exchange between the olivine of the melt impregnation and the olivine of the surrounding dunitic envelop, thereby raising the Mg# of the gabbroic olivine. As a result, this temperature estimate is most appropriately viewed as a maximum wall-rock temperature during emplacement. Owing to the small volume of spinel in this sample (shown in Pl. 1G), the olivine composition was most likely little affected by post-crystallization, high-temperature metamorphic exchange with spinel.

Overall, the evidence suggests that melt impregnation occurred as a sequence of events, with injection occurring as dikes, irregular melt concentrations, and small-scale, intergranular diffusive flow. Melt injection probably took place at temperatures well above 900°C and most likely approaching 1230°C, at depths below but near the base of the crust (4 to 10 km depth). These conditions are roughly similar to those proposed by Nicolas (1989) to explain the origin of similar melt-impregnated dunite bodies in ophiolitic harzburgite. Our results do not constrain whether melt impregnation took place under the ridge or slightly off-axis (both could have, in fact, taken place).

CONCLUSIONS

1. Depleted, residual harzburgites, now exposed by faulting at Hess Deep and sampled by coring during ODP Leg 147, were repeatedly impregnated with exotic N-MORB melt at depths of <10 km and temperatures substantially above 900°C and probably approaching 1230°C. Melt injection was accomplished by both diking and intergranular flow and occurred underneath or near the adjacent East Pacific Rise. This sequence of events is now represented by a generalized recovered stratigraphy of harzburgite-dunite-troctolite-gabbro-troctolite-dunite-harzburgite.

2. Ti contents in associated dunitic spinel are 10–20 times as high as in harzburgitic spinel, which indicates together with the mapped stratigraphy that the dunite recovered during Leg 147 is a relic of melt-wall rock interaction associated with melt impregnation. This process produced halos of dunite around residual bodies of gabbroic rock and possible residual dunite "trails" of melt migration through peridotite.

3. "Fresh" gabbroic spinel compositions vary widely $(TiO_2, 0.55\%-2.56\%; Mg#, 0.30-0.60)$, commonly within individual samples. This wide range in composition shows that the impregnating melts became locally evolved, with Ti-enriched melts produced on a centimeter scale by both crystal fractionation and local melt-rock interaction.

4. Ti in spinel is the element most easily analyzed by the electron microprobe for monitoring mass exchange of incompatible elements during melt–wall rock interaction, and was mobile during both melt impregnation and during the subsequent greenschist-zeolite metamorphic overprint.

5. Sharp increases in ZnO (to 0.75%) in the metamorphically altered rims of gabbro and troctolite spinels indicate that Zn-rich hydrothermal fluids may have been present during the alteration of melt-impregnated peridotites, with the gabbroic melt impregnations providing possible preferential conduits for fluid flow.

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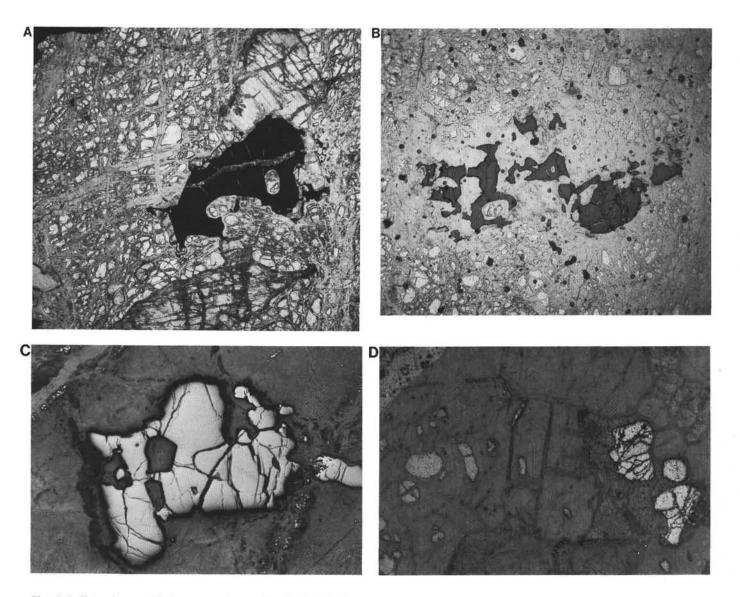


Plate 1. **A.** Photomicrograph (2.1 mm across, in transmitted light) of blocky, anhedral Cr-spinel in harzburgite Sample 147-895D-5R-1, 83–86 cm. Note also the presence of orthopyroxene porphyroclasts and smaller, anhedral grains of clinopyroxene attached to the spinel margin. Analyses of this spinel are given by transect C1 in Figure 6 and analysis C1-C in Table 2. **B.** Photomicrograph (2.8 mm across, in transmitted light) of holly-leaf Cr-spinels in harzburgite Sample 147-895C-4R-1, 10–12 cm. Note the much greater alteration of this sample than that shown in Figure 1. Analyses of these spinels are given by transects A1 and A2 in Figure 6 and analyses A1-C, A1-R, and A2-C in Table 2. **C.** Reflected light photomicrograph (3.5 mm across) of irregular, rounded Cr-spinel in dunite Sample 147-895D-9R-1, 101–105 cm. Analyses of this spinel are given by transects A1 and A2 in Figure 6 and analyses A1-C, A2-C, and A2-R in Table 2. **D.** Combination reflected- and transmitted-light photomicrograph (3.5 mm across) of irregular, rounded Cr-spinel in dunite Sample 147-895E-6R-3, 108–111 cm. Analyses of this spinel are given by transect B1 in Figure 6 and analysis B1-C in Table 2.

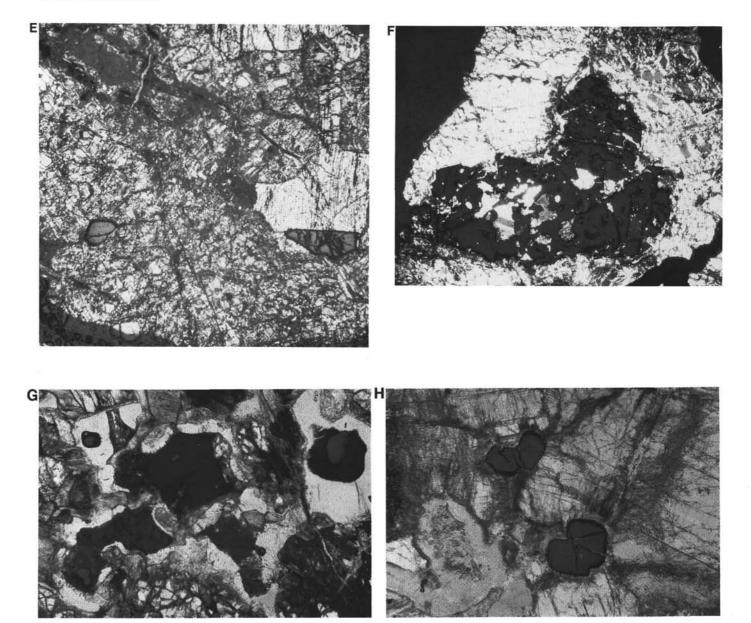


Plate 1 (continued). **E.** Combination reflected- and transmitted-light photomicrograph (1.9 mm across) of subhedral, rounded Cr-spinel in troctolite Sample 147-895C-4R-1, 63–68 cm. Melt segragations are represented by irregular patches of clinopyroxene surrounded by olivine. Spinel occurs as a grain on the edge of a melt segregation and as an isolated grain surrounded by olivine (represented by analyses 7-C and 8-C, respectively, in Table 2). **F.** Combination reflected- and transmitted-light photomicrograph (4.5 mm across) of skeletal Cr-spinel surrounded by olivine from troctolite Sample 147-895C-4R-2, 75–78 cm, and represented by analyses in transect C1 in Figure 6 and analyses C1-C and C2-C in Table 2. Note the plagioclase and clinopyroxene inclusions within the spinel; these are represented by analyses in Table 4. **G.** Combination reflected- and transmitted-light photomicrograph (1.75 mm across) of rounded, subhedral Cr-spinel inclusions in plagioclase that has been altered to chlorite, clays, and hydrogrossular (dark patches) and to serpentine and tremolite (surrounding lighter area) (Gillis, Mével, Allan, et al., 1993). These occur within a melt segregation of olivine gabbro Sample 147-895D-7R-2, 71–75 cm. The large spinel inclusion at the top of the photograph is represented by transect D1 of Figure 6 and by analysis D1-IPC in Table 2; the spinel inclusion at the lower right is represented by analysis 28-IPC in Table 2. Note the clusters of Fo₈₃₋₈₄ olivine grains (Table 3; light, heavily fractured grains). **H.** Combination reflected- and transmitted-light photomicrograph (1.75 mm across) of subhedral Cr-spinel inclusions in partially altered plagioclase within olivine gabbro Sample 147-895E-2R-1, 36–39 cm. The larger grain is represented by transect B1 in Figure 6 and by analyses B1-IPC and B1-IPR in Table 2.